# Nucleic-Acid Analogs with Restricted Conformational Flexibility in the SugarPhosphate Backbone ('Bicyclo-DNA') 

Part $7^{1}$ )

# Synthesis and Properties of Oligodeoxynucleotides Containing [( $\mathbf{3}^{\prime} \mathbf{S}, \mathbf{5}^{\prime} \mathbf{S}, \mathbf{6}^{\prime}$ R)- $\mathbf{6}^{\prime}$ -Amino- $2^{\prime}$-deoxy- $\mathbf{3}^{\prime}, 5^{\prime}$-ethano- $\beta$-d-ribofuranosyl]thymine ( $=\left(6^{\prime} R\right)$ - $6^{\prime}$-Amino-bicyclothymidine) 

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Dedicated to Prof. Dr. Frank Seela on the occasion of his 60th birthday


#### Abstract

We describe the synthesis of the acetamido- and trifluoroacetamido-functionalized bicyclo-thymidines $\mathbf{1 1}$ and 12, starting from the silyl enol ether 1, in 6 steps. These nucleosides were converted to the corresponding cyanoethyl phosphoramidite building blocks $\mathbf{1 6}$ and $\mathbf{1 7}$ and subsequently incorporated into the homothymidylate decamers 18-22. Upon deprotection of the oligomers, the trifluoroacetamido functions were cleaved, leaving behind a free amino function in the sugar-phosphate backbone that is protonated at neutral pH , giving rise to partially zwitterionic oligonucleotides. Pairing properties with the complementary DNA oligomer $\mathrm{d}\left(\mathrm{A}_{10}\right)$, as determined by UV/melting curves, revealed a slightly increased stability of the duplex $\mathrm{d}\left(\mathrm{A}_{10}\right) \cdot \mathbf{2 0}$, in which the decathymidylate sequence shows an alternating arrangement of natural thymidine and amino-bicyclothymidine residues, relative to the natural reference duplex. The dependence of $T_{\mathrm{m}}$ on the salt concentration of the medium is reduced in this case. Duplex destabilization occurs if the amino-bicyclo-thymidine residues are replaced by the charge-neutral acetamido-bicyclo-nucleosides (e.g., $\mathrm{d}\left(\mathrm{A}_{10}\right) \cdot \mathbf{2 2}$ ), most probably due to steric interference of the acetamido substituent with the backbone $\mathrm{P}-\mathrm{O}\left(5^{\prime}\right)$ bond.


1. Introduction. - The design and synthesis of new oligonucleotide analogs with improved pairing properties, enhanced biostability and bioavailability relative to natural DNA and RNA, is of importance with respect to potential applications as antisense agents in human therapy and biotechnology [2], in DNA or RNA diagnostics [3], and in materials and computer science [4]. Furthermore, such oligonucleotides serve as tools to chemically rationalize the supramolecular assembly of oligonucleotide single strands, thus contributing to the understanding of the interrelation between monomeric nucleoside structure and the association properties of oligomers thereof.

In this context, we recently prepared the conformationally constrained DNA analog bicyclo-DNA and explored its pairing properties in detail (Fig. 1) [5-8]. As the backbone torsion angle $\gamma$ in bicyclo-DNA, in contrast to natural A- and B-DNA, is preferentially located in the antiperiplanar (ap) and not in the synclinal ( $+s c$ ) arrangement, this provided a unique opportunity to explore the effect of a change of

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Bicycio-deoxynucleosides

(6'R)-6'-Amino-bicyclo-deoxynucleosides

Fig. 1. Structure of the bicyclo-deoxynucleosides and the (6'R)-6'-amino-bicyclo-deoxynucleosides
this particular torsion angle on the association mode of duplex formation [9]. This conformational study was further extended to oligodeoxynucleotides containing 5'-epi-bicyclo-deoxynucleosides, in which $\gamma$ is restricted to the $-a c /-s c$ conformational space [1].

In addition to probing the effect of structural preorganization on duplex formation, the carbocyclic ring in the bicyclo-deoxynucleosides is ideally suited for introducing functional groups into well-defined positions at the sugar-phosphate backbone. Such functional groups may serve various purposes, e.g., the adjustment of torsion angle $\gamma$ by changing the preferred conformation of the carbocyclic ring or the change of the electrostatic environment of the sugar-phosphate backbone. Furthermore, they may be useful as anchor points for the introduction of reporter molecules or chemically reactive groups, or to enhance the catalytic power of DNA enzymes [10]. Here we report on the synthesis of $\left(6^{\prime} R\right)$ - $6^{\prime}$-amino-bicyclo-thymidine, its incorporation into oligonucleotides, and a first assessment of the base-pairing properties of accordingly modified oligonucleotides.
2. Synthesis of the ( $\mathbf{6}^{\prime} \boldsymbol{R}$ )- $\mathbf{6}^{\prime}$-Amino-bicyclo-deoxysugar Unit. - Of primary interest to us were the $\left(6^{\prime} R\right)$ - $6^{\prime}$-amino-bicylo-deoxynucleosides, in which the amino function is attached to the concave $(\beta)$ side of the bicyclic ring system. The $(R)$-configuration was chosen for two reasons. From the cis relationship of the substituents at $\mathrm{C}\left(5^{\prime}\right)$ and $\mathrm{C}\left(6^{\prime}\right)$, we expected a conformational change of the carbocyclic ring, relative to the unsubstituted bicyclo-nucleosides, towards a conformation in which the $5^{\prime}$ - OH group is located in the pseudoaxial position. This orients torsion angle $\gamma$ into the $(+s c)$ range and thus mimicks closely the orientation observed in DNA duplexes of the A and B type. Furthermore, from model building, it appeared that attachment to the $\beta$ face orients the functional group into the direction of the major groove of a DNA duplex, while attachement to the convex $\alpha$ face would orient it straight away from the duplex into the solvent.

The synthesis started from the silyl enol ether 1 (Scheme 1), prepared and used previously in the synthesis of tricyclo-DNA [11]. Although the relative configuration at the anomeric center of $\mathbf{1}$ seemed to be of minor importance with respect to the stereochemical outcome of the envisaged transformations in the carbocyclic part of the molecule, both, the $\alpha$-D- and $\beta$-D-anomers were used separately and in parallel. While we anticipated that direct methods for the introduction of the amino substituent, e.g. via aziridination [12] of $\mathbf{1}$ would yield predominantly products with the amino substituent in $\alpha$-position, we preferred a classical procedure via hydroboration followed by exchange of the resulting OH substituent at $\mathrm{C}(7)$ by an $\mathrm{NH}_{2}$ substituent under inversion of configuration.

Scheme 1. Synthesis of the Amino-Functionalized Bicyclo-sugar 7


a) $\mathrm{BH}_{3} \cdot$ THF ( 3 equiv.), THF, $-78^{\circ} \rightarrow$ r.t., 31 h. b) Dess-Martin periodinane ( 2 equiv.), $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, r.t., $2 \mathrm{~h} . c$ ) $\left(\mathrm{MeONH}_{3}\right) \mathrm{Cl}\left(2\right.$ equiv. ), $\mathrm{NaOAc}(1$ equiv. $), 90 \% \mathrm{EtOH}$, r.t., 20 min . d) Raney-Ni, $\mathrm{EtOH} / \mathrm{H}_{2} \mathrm{O} /$ conc. $\mathrm{NH}_{3}$ soln. $10: 4: 0.8, \mathrm{H}_{2}(10 \mathrm{bar})$, r.t., $\left.12 \mathrm{~h} . e\right) \mathrm{CF}_{3} \mathrm{COOEt}, \mathrm{Et}_{3} \mathrm{~N}$ ( 0.2 equiv.), r.t., 1 h.

Hydroboration of $\mathbf{1}$ proceeded smoothly and produced the two diastereoisomeric diols $\mathbf{2}$ and $\mathbf{3}$ in yields $>90 \%$ and in ratios $\mathbf{3 / 2}$ of $3.5: 1$ in the $\alpha$-D series and $>10: 1$ in the $\beta$-d series ${ }^{2}$ ). The higher selectivity in the $\beta$-D series is most likely due to the MeO substituent that sterically obstructs the attack of the borane reagent from the $\beta$-face, thus pronouncing the already inherent preference for attack at the convex $\alpha$ face of the bicyclic system. While direct conversion of the secondary-alcohol function to an amino function in $\mathbf{3}$ via Mitsunobu reaction [13] failed in our hands, we had to adopt a redox procedure to introduce the desired functionality. Oxidation of $\mathbf{3}(\rightarrow \mathbf{4})$ followed by treatment with $O$-methylhydroxylamine gave the corresponding $(E) /(Z)$-oximes $\mathbf{5}$ in high yields in both the $\alpha$-D and $\beta$-D series. Subsequent hydrogenolysis with Raney-Ni

[^1]then produced in almost quantitative yield ${ }^{3}$ ) the corresponding amines, which were isolated as the corresponding trifluoracetates $6 / 7$. As expected, the reaction proceeded with considerable stereoselectivity with ratios of $\mathbf{6} / 7>6: 1$ in both the $\alpha$-D and $\beta$-D series.
3. X-Ray Analysis and Solution Structure of $\boldsymbol{\alpha - 7} \mathbf{7}^{4}$ ). - Suitable crystals of $\alpha-7$ were subjected to X-ray analysis (Fig. 2) ${ }^{5}$ ). The structure unequivocally supports the assignment of the relative configuration at the centers $\mathrm{C}(7)$ and $\mathrm{C}(8)$ and, furthermore, provides insight into the preferred conformational properties of the bicyclic sugar component. As in the unsubstituted bicyclo-deoxynucleosides, the furanose unit adopts an almost perfect $1^{\prime}$-exo conformation with a pseudorotation phase angle $P$ of $144.3^{\circ}$. Importantly, the carbocyclic ring exists in a conformation with the silyloxy substituent at $\mathrm{C}(8)$ in a pseudoaxial and the trifluoroacetamido substituent at $\mathrm{C}(7)$ in a pseudoequatorial position. Its conformation thus deviates substantially from that of the unsubstituted bicyclo-deoxynucleosides in which the corresponding OH substituents were shown to exist invariably in the pseudoequatorial orientation.

Translated into the structural description of the DNA backbone, the introduction of the amido substituent corrects torsion angle $\gamma$ from the (unnatural) ap-orientation (ca. $150^{\circ}$ ) to the (in A- and B-DNA naturally occurring) $s c$-orientation ( $77.2^{\circ}$ ), while torsion angle $\delta\left(135.5^{\circ}\right)$ remains largely unaffected with values as observed in B-DNA helices [15]. Analysis of the conformation of $\alpha-7$ in solution by NMR coupling constant analysis essentially confirms that occurring in the solid state [14].
4. Synthesis of Nucleosides and Building Blocks for DNA Synthesis. - With both anomers $\alpha$ - and $\beta-7$ in hand, we then approached the nucleosidation reaction (Scheme 2). Interestingly, the stereochemical outcome of the nucleosidation according to the Vorbrüggen procedure [16] was strongly dependent upon the configuration of the anomeric center of 7 and upon the temperature. In the $\alpha$-D series, ratios of nucleosides $\alpha / \beta-\mathbf{8}^{4}$ ) varied from $>6: 1$ at high temperature to $1: 1.7$ at low temperature. In the $\beta$-D series, no reaction took place at low temperatures, while mixtures $\alpha / \beta-\mathbf{8} 4: 1$ were obtained upon heating. The higher reactivity of $\alpha-7$ over $\beta-7$ is in accord with a
${ }^{3}$ ) During preliminary experiments for the reduction of $\mathbf{5}$ with Raney-Ni, we isolated amine $\mathbf{1 2}$ in yields of up to $25 \%$. This product most likely arises from reduction of the imine formed between acetaldehyde and the already formed amine. The source of the acetaldehyde is unknown but most probably arises from the solvent EtOH , from which it is produced in minor amounts by the catalyst. The formation of this by-product could be completely suppressed by addition of ammonia as a competing amine.


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${ }^{4}$ ) The descriptor $\alpha$ or $\beta$ preceding a key number refers to the configuration at the anomeric center.
${ }^{5}$ ) Crystallographic data (excluding structure factors) for $\alpha-7$ have been deposited with the Cambridge Crystallographic Data Centre as deposition No. CCDC-132022. Copies of the data can be obtained, free of charge, on application to the CCDC, 12 Union Road, Cambridge CB21EZ, UK (fax: +44 (1223)336033; e-mail: deposit@ccdc.can.ac.uk).


Fig. 2. ORTEP Plot of crystals of $\alpha-7$ : a) stereoscopic view ( $25 \%$ probability thermal ellipsoids) and b) view along the crystallographic a-axis
pronounced kinetic anomeric effect. The preferred formation of the $\beta$-D-nucleoside at low temperature is most likely the result of a kinetically controlled reaction, while formation of the $\alpha$-D-nucleoside at high temperature seems to occur under thermodynamic control ${ }^{6}$ ).

The further synthetic transformations into the building blocks for oligonucleotide synthesis were straightforward and are summarized in Scheme 3. Desilylation of the trifluoroacetyl-protected amine $\beta-\mathbf{8}$ afforded $\mathbf{1 2}$ in high yield, which could be tritylated to 14 and converted into the phosphoramidite 16 according to typical protocols in nucleotide chemistry. Moreover, the trifluoroacetyl group of $\mathbf{1 2}$ could be easily

[^2]Scheme 2. One-Pot Nucleosidation Reaction of Thymine with 7


Scheme 3. Preparation of Building Blocks $\mathbf{1 6}$ and $\mathbf{1 7}$ for Oligonucleotide Synthesis

a) MeOH , conc. $\mathrm{NH}_{3}$, r.t., 4 h. b) $\mathrm{Ac}_{2} \mathrm{O}$ (1 equiv.), $\mathrm{C}_{5} \mathrm{H}_{5} \mathrm{~N}, 0^{\circ}, 2$ h. c) $\mathrm{Et}_{3} \mathrm{~N} \cdot 3 \mathrm{HF}$ (1.3 equiv.), $\mathrm{C}_{5} \mathrm{H}_{5} \mathrm{~N}$, r.t., 24 h (11); $\mathrm{Bu}_{4} \mathrm{NF} \cdot 3 \mathrm{H}_{2} \mathrm{O}$ (8 equiv.), MeCN , AcOH (10 equiv.), r.t., 2 h (12). d) Conc. $\mathrm{NH}_{3}$ soln., MeOH , r.t. 45 min . e) $(\mathrm{MeO})_{2} \mathrm{TrOSO}_{2} \mathrm{CF}_{3}$ (1.5-2.0 equiv.), $\mathrm{C}_{5} \mathrm{H}_{5} \mathrm{~N}, 50^{\circ}, 16 \mathrm{~h} . f$ ) ${ }^{\mathrm{i}} \mathrm{Pr}_{2} \mathrm{NEt}$ (4 equiv.), $\mathrm{Cl}\left[\mathrm{P}\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CN}\right)\left(\mathrm{N}^{\mathrm{i}} \mathrm{Pr}_{2}\right)\right]$ ( 2 equiv.), MeCN, r.t., 2 h .
removed ( $\rightarrow \mathbf{1 3}$ ). A similar deprotection of the amino function in $\beta-\mathbf{8}(\rightarrow \mathbf{9})$ and subsequent transformation into the acetamide $\mathbf{1 0}$ provided access to the acetamido series. The corresponding phosphoramidite building block $\mathbf{1 7}$ was obtained, in analogy to the trifluoroacetyl series, from 10 via desilylation ( $\rightarrow \mathbf{1 1}$ ), tritylation ( $\rightarrow \mathbf{1 5}$ ), and phosphitylation. While the trifluoroacetyl group will be lost during standard oligonucleotide deprotection to produce a free, protonatable amino function, the acetyl group is stable under these conditions and thus leads to a largely isosteric but charge-neutral residue.
5. Synthesis of Oligonucleotides. - The synthesis of the decamers $\mathbf{1 8}$ - $\mathbf{2 2}$ containing $\left(6^{\prime} R\right)$ - $6^{\prime}$-amino- and ( $6^{\prime} R$ )-6'-acetamido-bicyclo-deoxythymidine (see Fig. 3) were carried out on a Pharmacia-Gene-Assembler-Plus ${ }^{\circledR}$ DNA synthesizer on the $1.3-\mu \mathrm{mol}$ scale in the 'trityl-off' mode (for details, see Exper. Part). Commercially available, thymidine-loaded controlled-pore glass (CPG) was used as the starting unit. Coupling yields, as monitored by the on-line trityl assay, amounted on average to $90-98 \%$, when the modified building blocks 16 and $\mathbf{1 7}$ were coupled to a natural thymidine unit. The same yields were observed for coupling of a natural thymidine unit to a $6^{\prime}$-amino-bicyclo-thymidine unit. The coupling of two consecutive amino-bicyclo-thymidine building blocks, however, was less efficient and proceeded with coupling yields per cycle of only $c a .60 \%$. This low yield, most likely arising from steric interference of the 6 'substituents, precluded the synthesis of completely modified decamers. After completion of chain assembly, the detachment from the solid support and deprotection was carried out by standard treatment with conc. $\mathrm{NH}_{3}$ solution.


Fig. 3. Sequences of the oligonucleotides 18-23

The crude oligomers were purified by anion-exchange HPLC and the purity of the collected fractions controlled by reversed-phase HPLC. As expected, the retention times during DEAE-HPLC of the decamers decreased with increasing content of amino-bicyclo-deoxynucleotide units, due to the reduced overall negative charge of the oligomers. The integrity of the oligomers was confirmed by matrix-assisted laser-desorption-ionization time-of-flight mass spectrometry (MALDI-TOF-MS) (Table 2, Exper. Part). The synthesis of sequence 23, containing unsubstituted bicyclodeoxythymidine units was described previously [6].

All sequences $\mathbf{1 8} \mathbf{- 2 2}$ were chemically stable (HPLC and MALDI-ToF-MS control) under the conditions used for recording UV/melting curves ( NaCl -containing aqueous buffer, $\mathrm{pH} 6-10,0-100^{\circ}$ ). No enhanced instability of the oligomers towards hydrolysis, due to the presence of the $6^{\prime}$-amino function was observed.
6. Pairing Properties. - Initial pairing properties with the DNA complement $\mathrm{d}\left(\mathrm{A}_{10}\right)$ were obtained from UV/melting-curve analysis in buffer solutions containing 150 mm

Table 1. $\mathrm{T}_{m}$ Values for Duplex Formation of Sequences 18-23 with d $\left(A_{10}\right)$. Buffer: 10 mm Na-cacodylate ( $\mathrm{pH} 6.0 / 7.0$ ), $10 \mathrm{~mm} \mathrm{Na} \mathrm{NPO}_{4}(\mathrm{pH} 8.0 / 9.0), 150 \mathrm{~mm} \mathrm{NaCl}$; $c$ (duplex) $3.7-4.0 \mu$ м.

|  | $\left.T_{\mathrm{m}}\left[{ }^{\circ}\right]^{\mathrm{a}}\right)$ |  |  |  |
| :---: | :---: | :---: | :---: | :---: |
|  | pH 6.0 | pH 7.0 | pH 8.0 | pH 9.0 |
| 18, d(T-T-T-T-T-t $\left.{ }^{+}-\mathrm{T}-\mathrm{T}-\mathrm{T}-\mathrm{T}\right)$ | 22.4 | 22.2 | 21.7 | 21.3 |
| $19 \mathrm{~d}\left(\mathrm{~T}-\mathrm{T}-\mathrm{T}-\mathrm{T}-\mathrm{t}^{+}-\mathrm{t}^{+}-\mathrm{T}-\mathrm{T}-\mathrm{T}-\mathrm{T}\right)$ | 18.9 | 19.0 | 18.4 | 18.2 |
| $20 \mathrm{~d}\left(\mathrm{t}^{+}-\mathrm{T}-\mathrm{t}^{+}-\mathrm{T}-\mathrm{t}^{+}-\mathrm{T}-\mathrm{t}^{+}-\mathrm{T}-\mathrm{t}^{+}-\mathrm{T}\right)$ | 27.5 | 27.3 | 26.1 | 24.6 |
| 21 d(T-T-T-T-T-t ${ }^{\text {Ac }}$-T-T-T-T) | 20.0 | 19.5 | 20.1 | 19.6 |
| $22 \mathrm{~d}\left(\mathrm{t}^{\text {Ac }}-\mathrm{T}-\mathrm{t}^{\text {Ac }}-\mathrm{T}-\mathrm{t}^{\text {Ac }}-\mathrm{T}-\mathrm{t}^{\text {Ac }}-\mathrm{T}-\mathrm{t}^{\text {Ac }}-\mathrm{T}\right)$ | 13.5 | 13.5 | 13.2 | 13.1 |
| $23 \mathrm{~d}(\mathrm{t}-\mathrm{T}-\mathrm{t}-\mathrm{T}-\mathrm{t}-\mathrm{T}-\mathrm{t}-\mathrm{T}-\mathrm{t}-\mathrm{T})$ | - | $21.0^{\text {b }}$ ) | - | - |

$\left.{ }^{\text {a }}\right) T_{\mathrm{m}} 23^{\circ}$ for $\mathrm{d}\left(\mathrm{A}_{10}\right) \cdot \mathrm{dT}\left({ }_{10}\right)$ under identical conditions at $\left.\mathrm{pH} 7.0 .{ }^{\mathrm{b}}\right)$ Taken from [6].
NaCl , and in the pH range $6.0-9.0 . T_{\mathrm{m}}$ Data were extracted from the melting curves and are summarized in Table 1.

While none of the oligonucleotides $\mathbf{1 8} \mathbf{- 2 3}$ showed any transition without the DNA complement, which excludes self-aggregation of the pyrimidine strands, all formed stable duplexes with $\left.\mathrm{d}\left(\mathrm{A}_{10}\right)^{7}\right)$. Compared to the natural reference duplex $\mathrm{d}\left(\mathrm{A}_{10}\right)$. $\mathrm{d}\left(\mathrm{T}_{10}\right)$, which under the given conditions melts with a $T_{\mathrm{m}}$ of $23^{\circ}$, the duplex $\mathbf{2 0}$. $\mathrm{d}\left(\mathrm{A}_{10}\right)$, with five charged residues $\left(\mathrm{t}^{+}\right)$spaced by five natural thymidine residues, displayed slightly enhanced $T_{\mathrm{m}}$ values. As expected, the $T_{\mathrm{m}}$ is dependent on pH and slightly decreases with increasing pH . Replacement of the positively charged $\mathrm{t}^{+}$by its acetylated, charge-neutral derivative $\mathrm{t}^{\mathrm{Ac}}$, as in the duplex $\mathbf{2 2} \cdot \mathrm{d}\left(\mathrm{A}_{10}\right)$, leads to a noticeable decrease of $T_{\mathrm{m}}$ compared to both, the all-DNA duplex $\mathrm{d}\left(\mathrm{A}_{10}\right) \cdot \mathrm{d}\left(\mathrm{T}_{10}\right)$ and the duplex $\mathrm{d}\left(\mathrm{A}_{10}\right) \cdot 23$, containing the underivatized bicyclo-thymidine unit t . As expected, duplex melting in the case of $\mathbf{2 2} \cdot \mathrm{d}\left(\mathrm{A}_{10}\right)$ is pH insensitive in the pH -range investigated. Duplexes containing the mono- and disubstituted pyrimidine sequences 18, 19, and 21 essentially follow the described overall properties.

Duplex formation in the case of $\mathbf{2 0} \cdot \mathrm{d}\left(\mathrm{A}_{10}\right)$ is less dependent on the salt concentration of the medium, compared to the duplex $\mathrm{d}\left(\mathrm{T}_{10}\right) \cdot \mathrm{d}\left(\mathrm{A}_{10}\right)$. In the range of $25-600 \mathrm{~mm} \mathrm{NaCl}$, a linear dependence of $T_{\mathrm{m}}$ from $\ln [\mathrm{NaCl}]$ was measured with slopes $\delta T_{\mathrm{m}} / \delta \ln [\mathrm{NaCl}]$ of $4.32 \mathrm{~K} \cdot \mathrm{M}^{-1}$ for $\mathbf{2 0} \cdot \mathrm{d}\left(\mathrm{A}_{10}\right)$ and $5.24 \mathrm{~K} \cdot \mathrm{M}^{-1}$ for $\mathrm{d}\left(\mathrm{T}_{10}\right) \cdot \mathrm{d}\left(\mathrm{A}_{10}\right)$. The lower dependence of $T_{\mathrm{m}}$ from the electrolyte concentration is again in agreement with the partially zwitterionic nature of the backbone of $\mathbf{2 0}$.
7. Discussion and Conclusions. - Zwitterionic oligonucleotides were prepared in the past either by derivatization of the pyrimidine $\mathrm{C}(5)$ position with a flexible aminohexyl chain [17][18] or by introducing a basic 2-(dimethylamino) ethylphosphoramidate group as the linking unit between nucleoside residues [19]. In both cases, a negligible salt-concentration dependence of duplex formation was reported. However, no benefit in terms of strength of duplex formation arose as a result of the modifications.

[^3]From the experiments presented here, we conclude that, indeed, the positive charge associated with the amino-bicyclo-nucleosides exhibits a slightly stabilizing effect at neutral pH , most probably by neutralizing intrastrand repulsion of two neighboring negatively charged phosphate units. However, this stabilizing electrostatic effect seems in part to be compromised by a repulsive steric interaction between the substituent at position $\mathrm{C}\left(6^{\prime}\right)$ of the bicyclo-nucleosides and the $5^{\prime}$-phosphate residues. This interpretation is corroborated by the relatively low $T_{\mathrm{m}}$ of duplexes containing the residues $\mathrm{t}^{\text {Ac }}$ (e.g., $\mathbf{2 2} \cdot \mathrm{d}\left(\mathrm{A}_{10}\right)$ ), in which charge contributions to stability are disentangled from structural contributions. Thus, geometrically constrained and suitably positioned cationic units in the DNA backbone can stabilize DNA duplexes and may do so even more effectively, if the charge-carrying group does not sterically interfere with the preferred backbone conformation.

Although the $6^{\prime}$-substituted $\left(6^{\prime} R\right)$-bicyclo-deoxynucleosides seem to prefer a conformation favoring the $+s c$ orientation of torsion angle $\gamma$ as deduced from the X-ray structure of the precursor $\alpha-7$, not enhanced but reduced affinity to the DNA target, compared to unmodified bicyclo-DNA is the result. This is evident from comparison of the $T_{\mathrm{m}}$ data of the duplexes $\mathbf{2 2} \cdot \mathrm{d}\left(\mathrm{A}_{10}\right)$ and $\mathbf{2 3} \cdot \mathrm{d}\left(\mathrm{A}_{10}\right)$. Model building suggests that this is due to repulsive nonbonding interactions between the substituent at $\mathrm{C}\left(6^{\prime}\right)$ and the adjacent $\mathrm{O}\left(5^{\prime}\right)-\mathrm{P}$ ester bond, perturbing the preferred conformation of the latter with respect to the B-DNA backbone structure (too many atoms in the sugarphosphate backbone(!), see [20]).

Given the relatively weak destabilizing effect upon replacement of one ( $\left.6^{\prime} R\right)-6^{\prime}$ -acetamido-bicyclo-deoxynucleoside residue for a thymidine residue within a decamer duplex, we envision the use of this scaffold, as well as of its ( $6{ }^{\prime} S$ )-isomer, in the future for the site-specific introduction of functional units into DNA double and triple helices.


#### Abstract

We gratefully acknowledge financial support from the Swiss National Science Foundation, from the Wander-Stiftung, Bern, and from Novartis AG, Basel. We thank the BENEFRI Small Molecule Crystallography Service directed by Prof. Helen Stoeckli-Evans for measuring the X-ray data set, Dr. Eugen Stulz for structure generation and refinement, and Dr. Verena Meyer for help with preparative HPLC.


## Experimental Part

General. Solvents for extraction: technical grade, distilled. Solvents for reactions: reagent grade, distilled over $\mathrm{CaH}_{2}$ ( $\mathrm{MeCN}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$, pyridine) or Na (THF). Reagents: if not otherwise stated, from Fluka, highest quality available. TLC: Merck $\operatorname{SiL} G-25 U V_{254}$; non-UV-visible compounds were stained by dipping the plate in a mixture of $\mathrm{EtOH}(180 \mathrm{ml})$, 4-methoxybenzaldehyde $(10 \mathrm{ml})$, conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$ soln. $(10 \mathrm{ml})$, and $\mathrm{AcOH}(2 \mathrm{ml})$, followed by heating with a heat gun. Flash column chromatography (FC): silica gel ( $30-60 \mu \mathrm{~m}$ ) from Baker. HPLC: Pharmacia-LKB-2249 gradient pump attached to an ABI-Kratos-Spectroflow-757-UV/VIS detector and a Tarkan- $W+W$ recorder $600 ; t_{\mathrm{R}}$ in min. UV/Melting curves: Varian-Cary-3E-UV/VIS spectrometer equipped with a temp. controller unit and connected to a Compaq-ProLinea-3/25-zs personal computer, temp. gradient $0.5^{\circ} / \mathrm{min}$; data-point collection in intervals of $c a .0 .3^{\circ}$; at $<20^{\circ}$, the cell compartment was flushed with $\mathrm{N}_{2}$ to avoid condensation of $\mathrm{H}_{2} \mathrm{O}$ on the UV cells; the transition temperature $T_{\mathrm{m}}$ was determined as the maximum of the first derivative of the melting curve using the software package Origin ${ }^{\mathrm{TM}}$ V5.0. M.p.: Büchi 510; uncorrected. Optical rotations: Perkin-Elmer-241 polarimeter; 10-mm cell. IR: Perkin Elmer FTIR 1600; $\tilde{v}$ in $\mathrm{cm}^{-1}$. NMR: Bruker AC-300, DRX500; $\delta$ in ppm, ${ }^{13} \mathrm{C}$ multiplicities from DEPT spectra, $J$ in Hz. EI-MS: Varian MAT CH-7A; ionizing voltage $70 \mathrm{eV} ; \mathrm{m} / \mathrm{z}$ (intensity in \%). Liquid secondary ion mass spectrometry (LSI-MS): Micromass Autospec $Q$, primary ions $\mathrm{Cs}^{+}(25 \mathrm{keV})$; matrix: dithioerythrol/dithio-DL-threitol. Matrix-assisted laser desorption ionization time-of-flight mass spectrometry (MALDI-TOF-MS) of oligonucleotides was performed as described in [21].
(1R,3S,5S,7R,8S)-8-\{[(tert-Butyl)dimethylsilyl]oxy\}-3-methoxy-2-oxabicyclo[3.3.0]octan-5,7-diol ( $\alpha$-2) and ( $1 \mathrm{R}, 3 \mathrm{~S}, 5 \mathrm{~S}, 7 \mathrm{~S}, 8 \mathrm{R}$ )-8-\{[(tert-Butyl)dimethylsilyl]oxy\}-3-methoxy-2-oxabicyclo[3.3.0]octan-5,7-diol ( $\alpha-\mathbf{3}$ ). A soln. of $\alpha-1(1.462 \mathrm{~g}, 5.11 \mathrm{mmol})$ in THF (abs.) was treated with $\mathrm{BH}_{3} \cdot$ THF ( $1 \mathrm{~m}, 3$ equiv.) at $-78^{\circ}$, allowed to warm up to r.t., and quenched after 31 h with sat. $\mathrm{NaHCO}_{3}$ soln. ( 26 ml ). A soln. of $\mathrm{KHSO}_{5}$ triple salt ( 12.55 g , 20.4 mmol ) in sat. $\mathrm{NaHCO}_{3}$ soln. ( 125 ml ) was added, and the mixture stirred for an additional 2 h . Extraction with $\mathrm{Et}_{2} \mathrm{O}(4 \times 200 \mathrm{ml})$ followed by drying the org. phase $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$ and evaporation afforded the crude products, which were separated by FC (hexane/AcOEt 5:1 $\rightarrow 0: 1$ ): $0.320 \mathrm{~g}(21 \%)$ of $\alpha-\mathbf{2}$ and $1.14 \mathrm{~g}(73 \%)$ of $\alpha$ 3, both as white crystals.

Data of $\alpha$-2: M.p. $87.0-87.5^{\circ}$. TLC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}, 7.5 \% \mathrm{MeOH}\right): R_{\mathrm{f}} 0.39$. IR $\left(\mathrm{CCl}_{4}\right): 3620 w, 3561 w, 2954 s, 2930 s$, $2858 m, 1472 w, 1463 w, 1447 w, 1389 w, 1362 w, 1316 w, 1293 w, 1253 m, 1197 m, 1145 m, 1125 m, 1100 s, 1070 s, 1028 \mathrm{v} s$, $1006 w, 983 w, 940 m, 863 m, 838 s .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.09,0.11\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.88\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.87$ $(d d, J=14.0,6.6,1 \mathrm{H}-\mathrm{C}(6)) ; 1.92(d, J=4.4, \mathrm{OH}) ; 2.12(d d, J=13.6,0.7,1 \mathrm{H}-\mathrm{C}(4)) ; 2.19(d d, J=13.8,3.9$, $1 \mathrm{H}-\mathrm{C}(4)) ; 2.29(d d, J=14.0,6.6,1 \mathrm{H}-\mathrm{C}(6)) ; 2.62(s, \mathrm{OH}) ; 3.33(s, \mathrm{MeO}) ; 3.78(d d, J=5.2,4.0, \mathrm{H}-\mathrm{C}(8))$; $4.02(d, J=3.7, \mathrm{H}-\mathrm{C}(1)) ; 4.13-4.21(m, \mathrm{H}-\mathrm{C}(7)) ; 5.07(d d, J=3.9,0.9, \mathrm{H}-\mathrm{C}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}(75 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right):-4.88,-4.70\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 18.03\left(s, \mathrm{Me}_{3} C \mathrm{Si}\right) ; 25.77\left(q, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 43.64,48.63(2 t, \mathrm{C}(4), \mathrm{C}(6)) ; 54.49$ $(q, \mathrm{MeO}) ; 78.29,82.69(2 d, \mathrm{C}(7), \mathrm{C}(8)) ; 84.34(s, \mathrm{C}(5)) ; 94.66(d, \mathrm{C}(1)) ; 106.79$ (d, C(3)). EI-MS: 273 (11, $\left.\left[^{M}-\mathrm{OMe}\right]^{+}\right), 131$ (100).

Data of $\alpha$-3: M.p. $95-97^{\circ}$. TLC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}+7.5 \% \mathrm{MeOH}\right): R_{\mathrm{f}} 0.32$. IR $\left(\mathrm{CHCl}_{3}\right): 3601 w, 3566 w$ (br.), $3003 m$, $2954 s, 2931 s, 2858 m, 1472 m, 1464 m, 1442 w, 1434 w, 1390 m, 1362 m, 1344 w, 1314 w, 1300 m, 1257 s, 1224 \mathrm{vs}, 1206 \mathrm{vs}$, $1142 s, 1098 s, 1080 s, 1050 m, 1020 m, 986 w, 949 m, 919 m, 882 m, 850 s, 839 s .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.09$ $\left(s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.88\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.70(d d, J=13.4,7.2,1 \mathrm{H}-\mathrm{C}(6)) ; 1.94(d d, J=13.8,4.2,1 \mathrm{H}-\mathrm{C}(4)) ; 2.11(d, J=$ 13.6, $1 \mathrm{H}-\mathrm{C}(4)) ; 2.21(d d, J=13.6,1 \mathrm{H}-\mathrm{C}(6)) ; 2.26(s, \mathrm{OH}) ; 2.91(s, \mathrm{OH}) ; 3.32(s, \mathrm{MeO}) ; 3.87-3.98$ $(m, \mathrm{H}-\mathrm{C}(7), \mathrm{H}-\mathrm{C}(8)) ; 4.16(d, J=5.5, \mathrm{H}-\mathrm{C}(1)) ; 5.06(d, J=4.4, \mathrm{H}-\mathrm{C}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ : $-4.90,-4.73\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 18.27\left(s, \mathrm{Me}_{3} C \mathrm{Ci}\right) ; 25.81\left(q, M e_{3} \mathrm{CSi}\right) ; 41.85,48.74(2 t, \mathrm{C}(4), \mathrm{C}(6)) ; 54.34(q, \mathrm{MeO})$; $76.85,77.80(2 d, \mathrm{C}(7), \mathrm{C}(8)) ; 83.75(s, \mathrm{C}(5)) ; 88.48(d, \mathrm{C}(1)) ; 106.51(d, \mathrm{C}(3)) . \mathrm{EI}-\mathrm{MS}: 273\left(5,[M-\mathrm{OMe}]^{+}\right)$, 215 (100).
(1R,3R,5S, 7S, 8R)-8-\{[(tert-Butyl)dimethylsilyl]oxy\}-3-methoxy-2-oxabicyclo[3.3.0]octan-5,7-diol ( $\beta$-3). From $\beta$ - $\mathbf{1}(588 \mathrm{mg}, 2.05 \mathrm{mmol})$, as described above. $\mathrm{FC}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}, 2.5 \% \mathrm{MeOH}\right)$ yielded $\beta-\mathbf{3}(566 \mathrm{mg}, 91 \%)$. White crystals. M.p. $124-125^{\circ}$. TLC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}, 5 \% \mathrm{MeOH}\right): R_{\mathrm{f}} 0.18$. IR $\left(\mathrm{CHCl}_{3}\right): 3602 m, 3423 w$ (br.), $2955 s$, $2931 s, 2900 m, 2858 m, 1558 w, 1472 m, 1463 m, 1450 w, 1389 w, 1362 w, 1310 m, 1252 s, 1159 s, 1109 s, 1068 s, 1034 m$, $1007 w, 988 m, 956 m, 877 m, 840 s, 800 v s .^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.09\left(s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.90\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.66$ $(d d d, J=13.2,9.1,1.1,1 \mathrm{H}-\mathrm{C}(6)) ; 1.93$ (br. $s, \mathrm{OH}) ; 2.09(d d, J=13.4,1.7,1 \mathrm{H}-\mathrm{C}(4)) ; 2.21$ (br. $s, \mathrm{OH}) ; 2.31$ $(d d d, J=13.6,5.8,1.6,1 \mathrm{H}-\mathrm{C}(4)) ; 2.41(d d, J=13.2,7.0,1 \mathrm{H}-\mathrm{C}(6)) ; 3.35(s, \mathrm{MeO}) ; 3.83(d d, J=7.7,5.9$, $\mathrm{H}-\mathrm{C}(8)) ; 4.04(d, J=5.5, \mathrm{H}-\mathrm{C}(1)) ; 4.23(d d, J=16.4,7.5, \mathrm{H}-\mathrm{C}(7)) ; 5.05(d d, J=5.7,1.6, \mathrm{H}-\mathrm{C}(3))$. ${ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right):-4.76$, $-4.62\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 18.22\left(s, \mathrm{Me}_{3} C \mathrm{Ci}\right) ; 25.84\left(q, M e_{3} \mathrm{CSi}\right) ; 44.18$, 49.15 $(2 t, \mathrm{C}(4), \mathrm{C}(6)) ; 54.56(q, \mathrm{MeO}) ; 76.04,79.05(2 d, \mathrm{C}(7), \mathrm{C}(8)) ; 83.75(s, \mathrm{C}(5)) ; 88.40(d, \mathrm{C}(1)) ; 105.69$ (d, C(3)). EI-MS: 273 ([ $M-\mathrm{OMe}^{+}, 3$ ), 215 (100).
(1R,3S,5R,8S)-8-\{[(tert-Butyl)dimethylsilyl]oxy\}-5-hydroxy-3-methoxy-2-oxabicyclo[3.3.0]octan-7-one ( $\alpha$ 4). To a soln. of $\alpha-\mathbf{3}(2.376 \mathrm{~g}, 7.81 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(36 \mathrm{ml})$, a soln. of Dess-Martin periodinane [22] ( 6.622 g , $15.58 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(32 \mathrm{ml})$ was added at r.t. After 2 h , the mixture was diluted with ${ }^{t} \mathrm{BuOMe}(200 \mathrm{ml})$ and extracted with sat. $\mathrm{NaHCO}_{3} / 20 \% \mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$ soln. 1:1 $(100 \mathrm{ml})$ and the org. phase dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated: crude $\alpha-4\left(2.515 \mathrm{~g}, 90 \%\right.$ pure by $\left.{ }^{1} \mathrm{H}-\mathrm{NMR}\right)$. Colorless oil that was directly introduced into the next step. Anal. data of a FC-purified sample (AcOEt/hexane $1: 4$ ): TLC $\left(5 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ : IR $\left(\mathrm{CHCl}_{3}\right): 3673 w, 3584 w$, $3509 w, 3425 w(\mathrm{sh})$, $3002 w, 2933 s, 2858 m, 1764 s, 1605 w, 1467 m, 1394 m, 1364 m, 1297 m, 1253 s, 1218 s, 1190 m$, $1145 m, 1076 s, 1019 m, 949 s, 842 s .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.10,0.13\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.89\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 2.15$ $(d d, J=13.8,1.7,1 \mathrm{H}-\mathrm{C}(4)) ; 2.25(d d, J=14.0,4.8,1 \mathrm{H}-\mathrm{C}(4)) ; 2.50(d, J=18.8,1 \mathrm{H}-\mathrm{C}(6)) ; 2.59(d d, J=18.8$, $1.5,1 \mathrm{H}-\mathrm{C}(6)) ; 2.94$ (br. $s, \mathrm{OH}) ; 3.38(s, \mathrm{MeO}) ; 4.27(d, J=5.9, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 4.37(d d, J=6.1,1.7$, $\mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 5.04(d d, J=5.0,1.7, \mathrm{H}-\mathrm{C}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right):-5.15,-4.76\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right)$; $18.45\left(s, \mathrm{Me}_{3} C \mathrm{Si}\right) ; 25.71\left(q, M e_{3} \mathrm{CSi}\right) ; 46.69(t, \mathrm{C}(6)) ; 48.06(t, \mathrm{C}(4)) ; 55.01(q, \mathrm{MeO}) ; 76.60,84.28$ ( $2 d, \mathrm{C}(1)$, $\mathrm{C}(8)) ; 80.38(s, \mathrm{C}(5)) ; 105.28(d, \mathrm{C}(3)) ; 210.57(s, \mathrm{C}(7)) . \mathrm{EI}-\mathrm{MS}: 271\left(2,[M-\mathrm{OMe}]^{+}\right), 213$ (100).
(1R,3R,5R,8S)-8-\{[(tert-Butyl)dimethylsilyl]oxy]-5-hydroxy-3-methoxy-2-oxabicyclo[3.3.0]octan-7-one ( $\beta-4$ ). From $\beta$-3 $(566 \mathrm{mg}, 1.86 \mathrm{mmol})$, as described above: crude $\beta-\mathbf{4}(740 \mathrm{mg})$. White solid that was used in the next step without further purification. Anal. data of a FC-purified sample $\left(7.5 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ : TLC $(7.5 \%$ $\left.\mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right): R_{\mathrm{f}} 0.54 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.08,0.13\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.91\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 2.21(d, J=12.5$, $1 \mathrm{H}-\mathrm{C}(4)) ; 2.29(d d d, J=12.5,4.6,1.3,1 \mathrm{H}-\mathrm{C}(4)) ; 2.59(d, J=18.4,1 \mathrm{H}-\mathrm{C}(6)) ; 2.72(d, J=18.4,1 \mathrm{H}-\mathrm{C}(6))$; $3.17(s, \mathrm{MeO}) ; 4.45(d, J=1.8, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 4.48(d, J=1.8, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 5.00(d, J=4.4$,
$\mathrm{H}-\mathrm{C}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right):-5.15,-4.69\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 18.36\left(s, \mathrm{Me}_{3} C \mathrm{Si}\right) ; 25.75\left(q, M e_{3} \mathrm{CSi}\right) ; 48.26$, $49.10(2 t, \mathrm{C}(4), \mathrm{C}(6)) ; 54.19(q, \mathrm{MeO}) ; 78.13,85.21(2 d, \mathrm{C}(1), \mathrm{C}(8)) ; 80.23(s, \mathrm{C}(5)) ; 104.40(d, \mathrm{C}(3)) ; 210.65$ ( $s, \mathrm{C}(7)$ ).
(1R,3S,5S,7E/Z,8R)-8-\{[(tert-Butyl)dimethylsilyl]oxy\}-5-hydroxy-3-methoxy-2-oxabicyclo[3.3.0]octan-6-one O-Methyloxime ( $\alpha-5$ ). A soln. of crude $\alpha-4(5.03 \mathrm{~g}, c a .15 \mathrm{mmol})$ in abs. EtOH $(92 \mathrm{ml})$ was treated with a $\left(\mathrm{MeONH}_{3}\right) \mathrm{Cl}$ soln. $(2.609 \mathrm{~g}, 31.24 \mathrm{mmol})$ and anh. $\mathrm{NaOAc}(1.281 \mathrm{~g}, 15.62 \mathrm{mmol})$ in $\mathrm{H}_{2} \mathrm{O}(11 \mathrm{ml})(\mathrm{pH}$ ca. 4$)$, and the mixture was stirred for 20 min at r.t. Dilution with ${ }^{t} \mathrm{BuOMe}(400 \mathrm{ml})$ followed by extraction with sat. $\mathrm{NaHCO}_{3}$ soln. ( 400 ml ), drying of the org. phase $\left(\mathrm{MgSO}_{4}\right)$, evaporation, and FC of the residual yellowish oil (AcOEt/hexane $4.5: 1 \rightarrow \mathrm{AcOEt}$ ) afforded $\alpha-5\left(4.626 \mathrm{~g}, 89 \% ;(E) /(Z) 5: 1\right.$ (tentative assignment) by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ) as a yellow solid. Anal. data are from separated isomers.

Data of $\alpha-5$ (apolar ( $E$ )-isomer (tentative)): TLC ( $2.5 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ): $R_{\mathrm{f}} 0.51$. IR $\left(\mathrm{CHCl}_{3}\right): 3516 w$ (br.), $3001 w, 2935 m, 2858 w, 1657 w, 1467 w, 1421 w, 1394 w, 1358 w, 1307 w, 1253 m, 1217 \mathrm{vs}, 1126 m, 1082 s, 1042 s, 945 m$, $901 m, 868 m, 837 m .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.06,0.07\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.85\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 2.06(d, J=13.2$, $1 \mathrm{H}-\mathrm{C}(4)) ; 2.27(d d, J=13.2,4.8,1 \mathrm{H}-\mathrm{C}(4)) ; 2.61(d, J=19.1,1 \mathrm{H}-\mathrm{C}(6)) ; 2.78(d, J=19.5,1 \mathrm{H}-\mathrm{C}(6)) ; 3.27$ $(s, \mathrm{OH}) ; 3.37(s, \mathrm{MeO}) ; 3.82(s, \mathrm{MeON}=\mathrm{C}) ; 4.25(d, J=5.1, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 4.36(d, J=5.5, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 5.11(d, J=4.8, \mathrm{H}-\mathrm{C}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 4.97,-4.75\left(q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 18.19\left(s, \mathrm{Me}_{3} C \mathrm{CSi}\right)$; $25.70\left(q, M e_{3} \mathrm{CSi}\right) ; 35.19,46.16(2 t, \mathrm{C}(4), \mathrm{C}(6)) ; 54.91$ ( $q, \mathrm{MeO}$ ); 61.81, 72.66, 89.93 ( $2 d, q, \mathrm{C}(1), \mathrm{C}(8)$, $M e \mathrm{ON}=\mathrm{C}) ; 84.48(s, \mathrm{C}(5)) ; 108.18(d, \mathrm{C}(3)) ; 160.37(s, \mathrm{C}(7))$. EI-MS: $331\left(1, M^{+}\right), 242(100)$.

Data of $\alpha-5$ (polar ( $Z$ )-isomer (tentative)): TLC ( $2.5 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ): $R_{\mathrm{f}} 0.35$. IR $\left(\mathrm{CHCl}_{3}\right): 3516 w$, $3001 w, 2934 m, 2858 w, 1466 w, 1428 w, 1393 w, 1358 w, 1311 w, 1250 m, 1217 \mathrm{vs}, 1133 m, 1082 s, 1046 s, 1008 w, 939 m$, $879 m, 839 m .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.03,0.06\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.83\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 2.09(d, J=13.2,1 \mathrm{H}-\mathrm{C}(4))$; $2.23(d d, J=13.2,4.4,1 \mathrm{H}-\mathrm{C}(4)) ; 2.52(d, J=18.4,1 \mathrm{H}-\mathrm{C}(6)) ; 2.88(d, J=18.0,1 \mathrm{H}-\mathrm{C}(6)) ; 3.35(s, \mathrm{OH}) ; 3.37$ $(s, \mathrm{MeO}) ; 3.80(s, \mathrm{MeON}=\mathrm{C}) ; 4.20(d, J=5.5, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 4.75(d t, J=0.7,5.5, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 5.12(d, J=4.4, \mathrm{H}-\mathrm{C}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): c a .-5\left(q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 18.17\left(s, \mathrm{Me}_{3} C \mathrm{Ci}\right) ; 25.65$ $\left(M e_{3} \mathrm{CSi}\right) ; 37.47,45.73(2 t, \mathrm{C}(4), \mathrm{C}(6)) ; 54.90(q, \mathrm{MeO}) ; 61.38,66.60(2 d, \mathrm{C}(1), \mathrm{C}(8)) ; 83.98(s, \mathrm{C}(5)) ; 90.66$ $(q, M e \mathrm{ON}=\mathrm{C}) ; 108.29(d, \mathrm{C}(3)) ; 160.60(s, \mathrm{C}(7))$. EI-MS: 331 (38, $\left.M^{+}\right), 210$ (100).
(1R,3R,5S,7E/Z,8R)-8-\{[(tert-Butyl)dimethylsilyl]oxy\}-5-hydroxy-3-methoxy-2-oxabicyclo[3.3.0]octan-6-one O-Methyloxime ( $\beta-\mathbf{5}$ ). From crude $\beta-\mathbf{4}(566 \mathrm{mg}, 1.86 \mathrm{mmol})$, as described above. $\mathrm{FC}(\mathrm{AcOEt} / \mathrm{hexane} 5: 1)$ gave $\beta-5\left(523 \mathrm{mg}, 85 \% ;(E) /(Z) 6: 1\right.$ (tentative assignment) by $\left.{ }^{1} \mathrm{H}-\mathrm{NMR}\right)$ as a colorless oil. Anal. data are from pure isomers:

Data of $\beta-5$ (apolar ( $E$ )-isomer (tentative)): TLC $\left(5 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right): R_{\mathrm{f}} 0.54$. IR $\left(\mathrm{CHCl}_{3}\right): 3600 w, 3418 w$ (br.), $3000 w, 2956 m, 2933 s, 2893 m, 2853 m, 1479 w, 1468 w, 1448 w, 1414 w, 1395 w, 1368 w, 1312 w, 1257 m, 1168 s$, $1110 m, 1052 s, 1024 w, 985 w, 971 w, 944 m, 888 m, 864 s, 844 s .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.11,0.12\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right)$; $0.91\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.75(s, \mathrm{OH}) ; 2.16-2.27(m, 2 \mathrm{H}-\mathrm{C}(4)) ; 2.41(d, J=18.8,1 \mathrm{H}-\mathrm{C}(6)) ; 3.15(d d, J=18.8$, 1.1, $1 \mathrm{H}-\mathrm{C}(6)) ; 3.29(s, \mathrm{MeO}) ; 3.83(s, \mathrm{MeON}=\mathrm{C}) ; 4.09(d, J=5.5, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 4.62(d d, J=5.5,1.1$, $\mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 5.13(d d, J=4.6,2.8, \mathrm{H}-\mathrm{C}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right):-4.93,-4.62\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right)$; $18.44\left(s, \mathrm{Me}_{3} C S i\right) ; 25.82\left(q, M e_{3} \mathrm{CSi}\right) ; 38.43,47.73(2 t, \mathrm{C}(4), \mathrm{C}(6)) ; 54.83(q, \mathrm{MeO}) ; 61.75,73.68(2 d, \mathrm{C}(1)$, $\mathrm{C}(8)) ; 88.18(q, M e \mathrm{ON}=\mathrm{C}) ; 83.25(s, \mathrm{C}(5)) ; 105.82(d, \mathrm{C}(3)) ; 160.30(s, \mathrm{C}(7)) . \mathrm{EI}-\mathrm{MS}: 332\left(2,[M+\mathrm{H}]^{+}\right), 331$ ( $3, M^{+}$), 242 (100).

Data of $\beta-5$ (polar ( $Z$ )-isomer (tentative) ): TLC $\left(5 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right): R_{\mathrm{f}} 0.43$. IR $\left(\mathrm{CHCl}_{3}\right): 3600 w, 3422 w$ (br.), $3000 w, 2967 s, 2933 s, 2904 m, 2862 m, 1476 m, 1463 m, 1445 w, 1390 w, 1362 w, 1312 w, 1296 w, 1257 m, 1129 s$, $1090 m, 1057 s, 1028 m, 985 w, 943 m, 887 m, 863 m, 844 m .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.09,0.11\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.89$ $\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.69(s, \mathrm{OH}) ; 2.18(d d, J=12.9,4.4, \mathrm{H}-\mathrm{C}(4)) ; 2.28(d d, J=12.9,5.1, \mathrm{H}-\mathrm{C}(4)) ; 2.41(d, J=17.3$, $1 \mathrm{H}-\mathrm{C}(6)) ; 3.18(d, J=17.7,1 \mathrm{H}-\mathrm{C}(6)) ; 3.38(s, \mathrm{MeO})) ; 3.81(s, \mathrm{MeON}=\mathrm{C}) ; 4.06(d, J=5.5, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 4.85(d, J=5.5, \mathrm{H}-\mathrm{C}(1)$ or $\mathrm{H}-\mathrm{C}(8)) ; 5.24(d d, J=5.5,4.4, \mathrm{H}-\mathrm{C}(3))$. EI-MS: $331\left(3, M^{+}\right), 89(100)$.
$\mathrm{N}-\{(1 \mathrm{R}, 3 \mathrm{~S}, 5 \mathrm{~S}, 7 \mathrm{~S}, 8 \mathrm{R})-8-\{[$ (tert-Butyl)dimethylsilyl]oxy\}-5-hydroxy-3-methoxy-2-oxabicyclo[3.3.0]oct-7-yl\}-2,2,2-trifluoroacetamide ( $\alpha-6$ ) and $\mathrm{N}-\{(1 \mathrm{R}, 3 \mathrm{~S}, 5 \mathrm{~S}, 7 \mathrm{R}, 8 \mathrm{R})-8-\{[($ tert-Butyl)dimethylsilyl]oxy\}-5-hydroxy-3-me-thoxy-2-oxabicyclo[3.3.0]oct-7-yll-2,2,2-trifluoroacetamide ( $\alpha-7$ ). In an autoclave, Raney- $\mathrm{Ni}(1.03 \mathrm{~g}$ ) was suspended in $\mathrm{H}_{2} \mathrm{O}(4 \mathrm{ml})$ and pretreated with $\mathrm{H}_{2}(10 \mathrm{bar})$ under heavy stirring. After 10 min , a degassed soln. of $\alpha-5((E) /(Z) ; 873 \mathrm{mg}, 2.64 \mathrm{mmol})$ in $\mathrm{EtOH}(10 \mathrm{ml})$, containing 0.8 ml of conc. $\mathrm{NH}_{3}$ soln. was carefully transferred to the autoclave, and the mixture was hydrogenated for 12 h at 10 bar. The mixture was filtered over Celite and the filtrate evaporated to give $796 \mathrm{mg}(99 \%)$ of the crude amines as a colorless oil. This oil was dissolved in $\mathrm{CF}_{3} \mathrm{COOEt}(1.55 \mathrm{ml})$ containing $70 \mu \mathrm{l}\left(0.2\right.$ equiv.) of $\mathrm{Et}_{3} \mathrm{~N}$, the soln. stirred for 1 h at r.t. and evaporated, and the residue separated by FC (AcOEt/hexane $2.5: 1 \rightarrow 1: 1): \alpha-6(144 \mathrm{mg}, 14 \%)$ as a colorless oil and $\alpha-7$ ( $888 \mathrm{mg}, 84 \%$ ) as a white solid.

Data of $\alpha-6$ : TLC ( $8 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ): $R_{\mathrm{f}} 0.50$. IR $\left(\mathrm{CHCl}_{3}\right): 3553 w, 3426 m, 2995 w, 2955 s, 2932 s, 2858 m$, $1723 \mathrm{vs}, 1541 \mathrm{~m}, 1471 \mathrm{~m}, 1442 w, 1386 w, 1347 w, 1296 w, 1280 w, 1252 s, 1189 \mathrm{~s}, 1170 \mathrm{vs}, 1131 s, 1105 s, 1055 m, 1021 m$, $1005 w, 947 m, 927 s, 886 m, 872 m, 842 s .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.07,0.09\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.86\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.73$ $(d d, J=13.6,6.6,1 \mathrm{H}-\mathrm{C}(6)) ; 2.01(d d, J=13.8,4.2,1 \mathrm{H}-\mathrm{C}(4)) ; 2.14(d, J=13.6,1 \mathrm{H}-\mathrm{C}(4)) ; 2.34(d d, J=14.0$, $5.9,1 \mathrm{H}-\mathrm{C}(6)) ; 3.06(s, \mathrm{OH}) ; 3.34(s, \mathrm{MeO}) ; 4.04-4.16(m, \mathrm{H}-\mathrm{C}(1), \mathrm{H}-\mathrm{C}(7), \mathrm{H}-\mathrm{C}(8)) ; 5.13$ ( $d, J=4.4$, $\mathrm{H}-\mathrm{C}(3)) ; 6.62(d, J=6.7, \mathrm{NH}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right):-5.18,-4.82\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 18.12\left(s, \mathrm{Me}_{3} C \mathrm{Ci}\right)$; $25.62\left(q, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 39.22,48.12(t, \mathrm{C}(4), \mathrm{C}(6)) ; 54.49(q, \mathrm{MeO}) ; 57.47,75.28,88.84$ (3d, C(1), C(7), C(8)); 85.16 $(s, \mathrm{C}(5)) ; 107.34(d, \mathrm{C}(3)) ; 115.68\left(q, J(\mathrm{C}, \mathrm{F})=288.1, \mathrm{CF}_{3}\right) ; 156.89(q, J(\mathrm{C}, \mathrm{F})=37.2, C \mathrm{OCF} 3)$. EI-MS: 368 (14, $M-\mathrm{OMe}]^{+}$), 310 (100).

Data of $\alpha$-7: M.p. $84-85^{\circ}$. TLC ( $8 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ): $R_{\mathrm{f}} 0.61$. IR ( $\mathrm{CHCl}_{3}$ ): $3531 w, 3427 m, 2998 w, 2956 m$, $2931 m, 2859 m, 1724 s, 1534 m, 1472 m, 1464 m, 1443 m, 1427 w, 1390 w, 1373 w, 1362 w, 1344 w, 1322 w, 1303 w, 1261 s$, $1173 s, 1092 s, 1043 s$, $966 m, 948 m, 875 m, 838 s .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.05,0.07\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.87$ $\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.89-2.01(m, 1 \mathrm{H}-\mathrm{C}(4), 1 \mathrm{H}-\mathrm{C}(6)) ; 2.15(d, J=13.6,1 \mathrm{H}-\mathrm{C}(4)) ; 2.23(d d, J=13.8,7.6$, $1 \mathrm{H}-\mathrm{C}(6)) ; 2.96(s, \mathrm{OH}) ; 3.34(s, \mathrm{MeO}) ; 4.14(t, J=4.0, \mathrm{H}-\mathrm{C}(8)) ; 4.19(d, J=4.4, \mathrm{H}-\mathrm{C}(1)) ; 4.53(d d d, J=$ $16.8,8.4,3.8, \mathrm{H}-\mathrm{C}(7)) ; 5.10(d, J=4.4, \mathrm{H}-\mathrm{C}(3)) ; 6.66(d, J=7.7, \mathrm{NH}) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right):-5.45$, $-4.43\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 18.10\left(s, \mathrm{Me}_{3} C \mathrm{Si}\right) ; 25.66\left(q, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 41.14,48.24(2 t, \mathrm{C}(4), \mathrm{C}(6)) ; 53.34(d, \mathrm{C}(7)) ; 54.65$ $(q, \mathrm{MeO}) ; 73.34(d, \mathrm{C}(8)) ; 84.95(s, \mathrm{C}(5)) ; 91.05(d, \mathrm{C}(1)) ; 107.96(d, \mathrm{C}(3)) ; 111.93\left(q, J(\mathrm{C}, \mathrm{F})=286, \mathrm{CF}_{3}\right)$; $156.24\left(q, J(\mathrm{C}, \mathrm{F})=37, \mathrm{CF}_{3} \mathrm{CO}\right)$. EI-MS: $368\left(11,[M-O M e]^{+}\right), 312(100)$.
$X$-Ray Analysis of $\alpha-7$ : Suitable crystals were obtained from $0.02 \mathrm{~m} \alpha-7$ in hexane at $-24^{\circ}$. Colorless transparent needles $(0.57 \times 0.27 \times 0.19 \mathrm{~mm}) ; \mathrm{C}_{16} \mathrm{H}_{28} \mathrm{~F}_{3} \mathrm{NO}_{5} \mathrm{Si}$; orthorhombic space group $P 212121$. Intensities were measured with a Stoe-AED-4-circle diffractometer ( $\mathrm{Mo}_{\alpha}, \lambda 0.71073 \AA$ ). Of the 2265 independent reflections $\left(\theta=2.03-25.47^{\circ}\right), 1645$ with $F>4 \sigma(F)$ were used in the refinement. The structure was solved using direct methods with SHELX-86 [23] and refined by full-matrix least-square procedures SHELXL-97. Non-Hatoms were refined anisotropically. The positions of all H -atoms were calculated and adjusted after every leastsquares cycle. The refinement converged at $R=0.0627, R w=0.1755$.
$\mathrm{N}-\{(1 \mathrm{R}, 3 \mathrm{R}, 5 \mathrm{~S}, 7 \mathrm{~S}, 8 \mathrm{R})-8-\{[($ tert-Butyl)dimethylsilyl]oxy\}-5-hydroxy-3-methoxy-2-oxabicyclo[3.3.0]oct-7-yl\}-2,2,2-trifluoroacetamide ( $\beta-6$ ) and $\mathrm{N}-\{(1 \mathrm{R}, 3 \mathrm{R}, 5 \mathrm{~S}, 7 \mathrm{R}, 8 \mathrm{R})-8-\{[$ (tert-Butyl)dimethylsilyl]oxy\}-5-hydroxy-3-me-thoxy-2-oxabicyclo[3.3.0]oct-7-yll-2,2,2-trifluoroacetamide ( $\beta-7$ ). Procedure not optimized. As described above, from $\beta-5((E) /(Z) ; 495 \mathrm{mg}, 1.50 \mathrm{mmol})$ by hydrogenation $\left(\mathrm{H}_{2} / 20 \mathrm{bar}\right)$ over $10 \% \mathrm{Pd} / \mathrm{C}(495 \mathrm{mg})$ in MeOH , followed by trifluoroacetylation. $\mathrm{FC}\left(2 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ yielded $\beta-6 / \beta-7(104 \mathrm{mg}, 17 \%$; ratio $1: 5$ by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ) besides $33 \%$ of nonconverted $\beta-5$.

Data of $\beta$-7: TLC $\left(10 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right): R_{\mathrm{f}} 0.64$. IR $\left(\mathrm{CHCl}_{3}\right): 3597 w, 3416 w, 3265 w(\mathrm{sh}), 3101 w, 2954 m$, 2931s, 2901m, 2858m, 1720vs, 1548m, 1472m, 1464m, 1448m, 1385m, 1362m, 1314m, 1282m, 1256s, 1166vs, $1118 m, 1103 m, 1088 m, 1071 m, 1031 m, 1006 w, 978 m, 959 m, 940 m, 922 m, 865 m, 839 s .{ }^{1} \mathrm{H}-\mathrm{NMR}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): 0.06,0.10\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.87\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 2.13(s, \mathrm{OH}) ; 2.21-2.26\left(m, 3 \mathrm{H}, \mathrm{CH}_{2}(4), \mathrm{CH}_{2}(6)\right) ; 2.45$ $\left(d d, J=14.0,6.3,1 \mathrm{H}, \mathrm{CH}_{2}(4), \mathrm{CH}_{2}(6)\right) ; 3.41(s, \mathrm{MeO}) ; 4.05(d, J=4.8, \mathrm{H}-\mathrm{C}(1)) ; 4.30(d d, J=6.4,5.0$, $\mathrm{H}-\mathrm{C}(8)) ; 4.56(m, \mathrm{H}-\mathrm{C}(7)) ; 5.17(d d, J=6.6,2.2, \mathrm{H}-\mathrm{C}(3)) ; 8.08(d, J=8.1, \mathrm{NH}) .{ }^{13} \mathrm{C}-\mathrm{NMR}(75 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right):-5.27,-5.05\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 18.13\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 25.61\left(q, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 43.87,49.03(2 t, \mathrm{C}(4), \mathrm{C}(6)) ; 53.03$, $56.22(d+q, \mathrm{C}(7), \mathrm{MeO})) ; 73.09,90.11(2 d, \mathrm{C}(1), \mathrm{C}(8)) ; 84.31(s, \mathrm{C}(5)) ; 107.39(d, \mathrm{C}(3)) ; 117.95\left(q, \mathrm{CF}_{3}\right)$; $156.75\left(q, \mathrm{CF}_{3} \mathrm{CO}\right)$.
(3'S,5'R, $\left.6^{\prime} \mathrm{R}\right)-1-\left\{5^{\prime}-\mathrm{O}-\left[\left(\right.\right.\right.$ tert-Butyl)dimethylsilyl]-2'-deoxy- $6^{\prime}$-(trifluoroacetamido)-3',5'-ethano- $\alpha / \beta$-D-ribofuranosylfthymine $(\alpha / \beta-8)$. A mixture of $\alpha-7(1.000 \mathrm{~g}, 2.51 \mathrm{mmol})$ and dry thymine $(631 \mathrm{mg}, 5.00 \mathrm{mmol})$ in abs. $\mathrm{MeCN}(25 \mathrm{ml})$ was cooled to $3^{\circ}$ and treated with $\mathrm{N}, \mathrm{O}$-bis(trimethylsilyl) acetamide (BSA; $3.06 \mathrm{ml}, 1.25 \mathrm{mmol}$ ) followed by $\mathrm{Me}_{3} \mathrm{SiCl}(158 \mu \mathrm{l}, 1.25 \mathrm{mmol})$. After $90 \mathrm{~min}(\rightarrow$ clear soln. $), \mathrm{Me}_{3} \mathrm{SiOSO}_{2} \mathrm{CF}_{3}(1.82 \mathrm{ml}, 10.03 \mathrm{mmol})$ was added and the mixture stirred for 20 h at $3^{\circ}$. Then the mixture was treated with $1 \mathrm{~m} \mathrm{HCl}(25 \mathrm{ml})$ for 5 min , diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(150 \mathrm{ml})$, and extracted with sat. $\mathrm{NaHCO}_{3}$ soln. The org. phases were dried $\left(\mathrm{MgSO}_{4}\right)$, and evaporated and the crude nucleosides purified by $\mathrm{FC}\left(0-3 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give $\alpha / \beta-8(1.110 \mathrm{~g}, 90 \%$; ratio $1: 1.5$ by ${ }^{1} \mathrm{H}-\mathrm{NMR}$ ) as a white foam. Separation of the anomers was effected by prep. HPLC (LiChrosorb SI 60, $7 \mu, 23 \times 250 \mathrm{~mm}$; hexane $\left./ i \operatorname{PrOH} 8: 2 ; t_{\mathrm{R}} 22.5(\beta-\mathbf{8}), 27.5(\alpha-\mathbf{8})\right)$ to give anal. pure $\beta-\mathbf{8}(577 \mathrm{mg}, 47 \%)$ and $\alpha-\mathbf{8}$ ( $332 \mathrm{mg}, 27 \%$ ), both as white foams.

Data of $\alpha-8$ : TLC (hexane/iPrOH 8:2): $R_{\mathrm{f}} 0.35$. IR $\left(\mathrm{CHCl}_{3}\right): 3426 m, 2956 m, 2932 m, 2900 w, 2860 m, 1691 \mathrm{vs}$, $1535 m, 1472 m, 1443 m, 1427 w, 1411 w, 1364 m, 1264 s, 1237 s, 1229 s, 1212 s, 1200 s, 1173 s, 1104 s, 1051 m, 1006 m$, $967 m, 872 m, 837 s .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.07,0.12\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.90\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.84(s, \mathrm{Me}-\mathrm{C}(5)) ; 2.06$ $\left(d d J=13.6,9.9,1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.38\left(d d, J=13.6,7.7,1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.49-2.63\left(m, 2 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 4.18(t, J=3.9$, $\left.\mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 4.33$ (br. $\left.s, \mathrm{OH}\right) ; 4.53-4.63\left(m, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right) ; 4.58\left(d, J=4.0, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right) ; 6.06(d d, J=6.8,2.8$, $\left.\mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 6.66\left(d, J=8.1, \mathrm{CF}_{3} \mathrm{CONH}\right) ; 7.32(s, \mathrm{H}-\mathrm{C}(6)) ; 9.76$ (br. $\left.s, \mathrm{H}-\mathrm{N}(3)\right) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ :
$-5.44,-4.47\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 12.40(q, M e-\mathrm{C}(5)) ; 18.00\left(s, \mathrm{Me}_{3} C \mathrm{Si}\right) ; 25.68\left(q, M e_{3} \mathrm{CSi}\right) ; 43.40,47.99\left(2 t, \mathrm{C}\left(2^{\prime}\right)\right.$, $\left.\mathrm{C}\left(7^{\prime}\right)\right) ; 52.05\left(d, \mathrm{C}\left(6^{\prime}\right)\right) ; 73.47\left(d, \mathrm{C}\left(5^{\prime}\right)\right) ; 84.34\left(s, \mathrm{C}\left(3^{\prime}\right)\right) ; 91.40\left(\mathrm{C}\left(1^{\prime}\right)\right) ; 93.32\left(d, \mathrm{C}\left(4^{\prime}\right)\right) ; 109.95(s, \mathrm{C}(5)) ; 115.67$ $\left(q, J(\mathrm{C}, \mathrm{F})=288, \mathrm{CF}_{3}\right) ; 137.58(s, \mathrm{C}(6)) ; 150.56(s, \mathrm{C}(2)) ; 156.52\left(q, J(\mathrm{C}, \mathrm{F})=37, \mathrm{COCF}_{3}\right) ; 164.53(s, \mathrm{C}(4))$. LSIMS: $646.2\left(2,[M+\text { matrix }]^{+}\right), 532.1\left(3,[M+\mathrm{K}]^{+}\right), 516.2\left(2,[M+\mathrm{Na}]^{+}\right), 494.2\left(39,[M+1]^{+}\right), 368.2(100)$.

Data of $\beta-\mathbf{8}$ : TLC (hexane/iPrOH $8: 2$ ): $R_{\mathrm{f}} 0.27$. IR $\left(\mathrm{CHCl}_{3}\right): 3596 w, 3397 m, 2957 m, 2932 m, 2899 w, 2860 w$, $1693 \mathrm{vs}, 1532 \mathrm{~m}, 1472 \mathrm{~m}, 1442 \mathrm{w}, 1427 \mathrm{w}, 1414 \mathrm{w}, 1371 \mathrm{~m}, 1318 \mathrm{w}, 1281 \mathrm{~m}, 1261 \mathrm{~s}, 1238 \mathrm{~s}, 1213 \mathrm{~s}, 1199 \mathrm{~s}, 1175 \mathrm{~s}, 1101 \mathrm{~s}$, $1056 m, 1012 m, 977 m, 931 m, 910 m, 839 s .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.07\left(s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.89\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.87$ $(s, \mathrm{Me}-\mathrm{C}(5)) ; 1.96\left(d d, J=13.2, \quad 9.9,1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 2.16-2.30\left(m, 2 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.55 \quad(d d, J=13.6, \quad 5.2$, $\left.1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 4.10\left(d, J=5.2, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right) ; 4.38\left(t, J=5.0, \quad \mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 4.44-4.50\left(m, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right) \mathrm{OH}\right) ; 6.32$ $\left(d d, J=9.6,5.5, \mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 6.98\left(d, J=6.6, \mathrm{CF}_{3} \mathrm{CONH}\right) ; 7.21(s, \mathrm{H}-\mathrm{C}(6)) ; 10.19(s, \mathrm{H}-\mathrm{N}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}$ ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $-5.40,-4.52\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 12.14(q, \mathrm{Me}-\mathrm{C}(5)) ; 18.13\left(s, \mathrm{Me}_{3} C \mathrm{Si}\right) ; 25.72\left(q, M e_{3} \mathrm{CSi}\right) ; 42.36$, $46.58\left(2 t, \mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(7^{\prime}\right)\right) ; 54.20\left(d, \mathrm{C}\left(6^{\prime}\right)\right) ; 71.80\left(d, \mathrm{C}\left(5^{\prime}\right)\right) ; 83.75\left(s, \mathrm{C}\left(3^{\prime}\right)\right) ; 85.49\left(d, \mathrm{C}\left(1^{\prime}\right)\right) ; 88.99\left(d, \mathrm{C}\left(4^{\prime}\right)\right)$; $111.82(s, \mathrm{C}(5)) ; 115.72\left(q, J(\mathrm{C}, \mathrm{F})=288, \mathrm{CF}_{3}\right) ; 134.91(d, \mathrm{C}(6)) ; 150.87(s, \mathrm{C}(2)) ; 156.74(q, J(\mathrm{C}, \mathrm{F})=37$, $\left.C \mathrm{COCF}_{3}\right) ; 163.87(s, \mathrm{C}(4))$. LSI-MS: $646.1\left(1,[M+\text { matrix }]^{+}\right), 532.1\left(4,[M+\mathrm{K}]^{+}\right), 516.1\left(2,[M+\mathrm{Na}]^{+}\right), 494.1$ (39, $\left.[M+1]^{+}\right), 368.1$ (100).
(3'S,5'R, $6^{\prime} \mathrm{R}$ )-1-\{6'-Amino-5'-O-[(tert-butyl)dimethylsilyl]-2'-deoxy-3',5'-ethano- $\alpha / \beta$-D-ribofuranosyl\}thymine (9). To a soln. of $\beta-8(640 \mathrm{mg}, 1.3 \mathrm{mmol})$ in $\mathrm{MeOH}(1.3 \mathrm{ml})$, conc. $\mathrm{NH}_{3}$ soln. $(26 \mathrm{ml})$ was added. The resulting mixture was stirred for 4 h at r.t. and subsequently evaporated. $\mathrm{CC}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} / \mathrm{NH}_{3}\right.$ soln. $\left.50: 5: 3\right)$ gave 9 ( 472 mg , $91 \%$ ). Slightly yellow oil. TLC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} / \mathrm{NH}_{3}\right.$ soln. $\left.18: 2: 1\right): R_{\mathrm{f}} 0.43$. UV ( MeOH ): 264 (9930). IR (KBr): $3419 s$ (br.), $3198 s$ (br), $3062 s$ (br.), 2954s, 2930s, 2898s, 2857s, 1698vs, 1686vs, 1472m, 1287w, 1262w, $1152 w, 836 w, 780 w .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 0.07,0.10\left(2 s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.91\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.84-1.99$ $\left(m, 2 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 1.88(s, \mathrm{Me}-\mathrm{C}(5)) ; 2.33\left(d d, J=13.1,9.6,1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 2.41\left(d d, J=13.2,5.5,1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right)$; $3.62\left(d d, J=10.3,4.3, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right) ; 4.03\left(d, J=5.7, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right) ; 4.11-4.14\left(m, \mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 6.36(d d, J=9.6,5.3$, $\left.\mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 7.79(d, J=1.1, \mathrm{H}-\mathrm{C}(6)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right):-5.00,-4.45\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 12.21$ $(q, \mathrm{Me}-\mathrm{C}(5)) ; 18.26\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 25.91\left(q, M e_{3} \mathrm{CSi}\right) ; 43.74,46.72\left(2 t, \mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(7^{\prime}\right)\right) ; 55.70\left(d, \mathrm{C}\left(6^{\prime}\right)\right) ; 73.60$ $\left(d, \mathrm{C}\left(5^{\prime}\right)\right) ; 84.55\left(s, \mathrm{C}\left(3^{\prime}\right)\right) ; 87.03\left(d, \mathrm{C}\left(1^{\prime}\right)\right) ; 89.21\left(d, \mathrm{C}\left(4^{\prime}\right)\right) ; 110.92(s, \mathrm{C}(5)) ; 137.79(d, \mathrm{C}(6)) ; 151.13(s, \mathrm{C}(2))$; 164.27 ( $s, \mathrm{C}(4))$. HR-LSI-MS $\left(\mathrm{C}_{18} \mathrm{H}_{32} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{Si}\right)$ : 398.2111 (calc. 398.2109).
( $3^{\prime} \mathrm{S}, 5^{\prime} \mathrm{R}, 6^{\prime} \mathrm{R}$ )-1-\{6'-Acetamido-5'-O-[(tert-butyl)dimethylsilyl]-2'-deoxy-3',5'-ethano- $\beta$-D-ribofuranosyl\}thymine (10). To a soln. of $\mathbf{9}(396 \mathrm{mg}, 0.99 \mathrm{mmol})$ in pyridine $(3.5 \mathrm{ml}), \mathrm{Ac}_{2} \mathrm{O}(102 \mu \mathrm{l}, 1 \mathrm{mmol})$ was added at $0^{\circ}$, and the mixture was stirred for 2 h . Dilution with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(50 \mathrm{ml})$, extraction with sat. $\mathrm{NaHCO}_{3}$ soln. $(50 \mathrm{ml})$, and evaporation of the dried $\left(\mathrm{MgSO}_{4}\right)$ org. phase, followed by $\mathrm{FC}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 100: 8\right)$ yielded $\mathbf{1 0}(387 \mathrm{mg}$, $88 \%$ ). White foam. TLC ( $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 100: 8\right): R_{\mathrm{f}} 0.48$. UV (MeOH): 262 (9480). IR (KBr): $3368 m$ (br.), $3068 w$ (br.), 2954m, 2930m, 2892w, 2856m, 1700s, 1540w, 1472m, 1374m, 1284m, 1260m, 1154m, 1046m, 980w, $883 w, 836 m, 779 m .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): 0.11,0.13\left(s, \mathrm{Me}_{2} \mathrm{Si}\right) ; 0.98\left(s, \mathrm{Me}_{3} \mathrm{CSi}\right) ; 1.93(s, \mathrm{Me}-\mathrm{C}(5))$; $2.01(s, \mathrm{MeCO}) ; 2.01-2.07\left(m, 1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.16-2.26\left(m, 1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right), 1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.36(d d, J=13.1,5.2$, $\left.1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 4.06\left(d, J=4.8, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right) ; 4.40\left(t, J=4.4, \mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 4.44-4.49\left(m, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right) ; 6.23(d d, J=10.0$, $\left.5.2, \mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 7.53(d, J=1.1, \mathrm{H}-\mathrm{C}(6)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right):-4.48,-3.75\left(2 q, \mathrm{Me}_{2} \mathrm{Si}\right) ; 12.77$ $(q, M e-\mathrm{C}(5)) ; 19.78\left(s, \mathrm{Me}_{3} C S i\right) ; 23.09(q, M e \mathrm{CO}) ; 26.91\left(q, M e_{3} \mathrm{CSi}\right) ; 43.42$, $46.47\left(2 t, \mathrm{C}\left(2^{\prime}\right), \mathrm{C}^{\prime}\left(7^{\prime}\right)\right) ; 55.13$ ( $\left.d, \mathrm{C}\left(6^{\prime}\right)\right) ; 73.72\left(d, \mathrm{C}\left(5^{\prime}\right)\right) ; 84.27\left(s, \mathrm{C}\left(3^{\prime}\right)\right) ; 88.63\left(d, \mathrm{C}\left(1^{\prime}\right)\right) ; 91.42\left(d, \mathrm{C}\left(4^{\prime}\right)\right) ; 111.78(s, \mathrm{C}(5)) ; 138.75(d, \mathrm{C}(6))$; $152.52(s, C(2)) ; 166.45(s, C(4)) ; 173.06(s, \mathrm{MeCO})$. HR-LSI-MS $\left(\mathrm{C}_{20} \mathrm{H}_{34} \mathrm{~N}_{3} \mathrm{O}_{6} \mathrm{Si}\right): 440.2217$ (calc. 440.2201$)$.
(3'S,5'R,6'R)-1-(6'-Acetamido-2'-deoxy-3',5'-ethano- $\beta$-D-ribofuranosyl)thymine (11). To a soln. of $\mathbf{1 0}$ $(305 \mathrm{mg}, 0.69 \mathrm{mmol})$ in pyridine $(1.5 \mathrm{ml}), \mathrm{Et}_{3} \mathrm{~N} \cdot 3 \mathrm{HF}(160 \mu \mathrm{l}, 0.98 \mathrm{mmol})$ was added and the mixture stirred. After 24 h at r.t., excess solid $\left(\mathrm{NH}_{4}\right)_{2} \mathrm{CO}_{2}$ was added, the mixture was filtered, and the filtrate evaporated. FC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 100: 8\right)$ gave $11(215 \mathrm{mg}, 96 \%)$. White foam. TLC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 100: 8\right): R_{\mathrm{f}} 0.11$. UV ( $\left.\mathrm{H}_{2} \mathrm{O}\right)$ : 264 (9370). IR (KBr): $3384 s$ (br.), $3066 m, 2930 m, 2820 w, 1700 s, 1684 s, 1654 m, 1540 m, 1472 m, 1374 m, 1288 m$, $1264 m, 1144 m, 1056 m, 940 w, 814 w, 782 w, 668 w, 610 w, 560 w .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): 1.92(d, J=1.3$, $\mathrm{Me}-\mathrm{C}(5)) ; 2.02(s, \mathrm{MeCO}) ; 2.07-2.23\left(m, 1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right), 2 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.40\left(d d, J=13.0,5.2,1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 4.15$ $\left(t, J=3.8, \mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 4.21\left(d, J=4.2, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right) ; 4.52\left(d d d, J=11.0,8.6,3.3, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right) ; 6.44(d d, J=10.1,5.2$, $\left.\mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 8.20(d, J=1.1, \mathrm{H}-\mathrm{C}(6)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 12.81(q, \mathrm{Me}-\mathrm{C}(5)) ; 22.84(q, \mathrm{MeCO})$; 43.21, $47.76\left(2 t, \mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(7^{\prime}\right)\right) ; 54.05\left(d, \mathrm{C}\left(6^{\prime}\right)\right) ; 84.27\left(s, \mathrm{C}\left(3^{\prime}\right)\right) ; 89.30,92.72\left(2 d, \mathrm{C}\left(1^{\prime}\right), \mathrm{C}\left(4^{\prime}\right)\right) ; 111.76(s, \mathrm{C}(5))$; 139.07 ( $d, \mathrm{C}(6)$ ); $152.60(s, \mathrm{C}(2)) ; 166.67(s, \mathrm{C}(4)) ; 173.22(s, \mathrm{MeCO})$. HR-LSI-MS $\left(\mathrm{C}_{14} \mathrm{H}_{20} \mathrm{~N}_{3} \mathrm{O}_{6}\right): 326.1352$ (calc. 326.1358).
(3'S,5'R,6'R)-1-[2'-Deoxy-6'-(trifluoroacetamido)-3',5'-ethano- $\beta$-D-ribofuranosyl]thymine (12). To a soln. of $\beta-\mathbf{8}(993 \mathrm{mg}, 2.00 \mathrm{mmol})$ in $\mathrm{MeCN}(10 \mathrm{ml})$, a soln. of $\mathrm{Bu}_{4} \mathrm{NF} \cdot 3 \mathrm{H}_{2} \mathrm{O}(5.05 \mathrm{~g}, 16.02 \mathrm{mmol})$ in $\mathrm{MeCN}(10 \mathrm{ml})$ and $\mathrm{AcOH}(1.15 \mathrm{ml}, 20.01 \mathrm{mmol})$ was added, and the resulting mixture was stirred for 2 h at r.t. Dilution with AcOEt ( 100 ml ), extraction with sat. $\mathrm{NaHCO}_{3}$ soln., followed by drying of the org. phase $\left(\mathrm{MgSO}_{4}\right)$ gave crude
product that was purified by FC (AcOEt/hexane $3: 1$ ): $\mathbf{1 2}$ ( $719 \mathrm{mg}, 95 \%$ ). White foam. TLC ( $8 \% \mathrm{MeOH} /$ $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ): $R_{\mathrm{f}} 0.19$. UV ( $\mathrm{H}_{2} \mathrm{O}$ ): 265 (8910). IR ( KBr ): $3408 s$ (br.), $3000 s$ (sh), $3072 m, 2954 w, 2830 w, 1698 \mathrm{vs}$, $1562 m, 1552 m, 1536 w, 1475 m, 1408 w, 1376 m, 1289 s, 1266 s, 1215 s, 1188 s, 1158 s, 1107 m, 1054 m, 1010 w, 968 w$, $944 m, 882 w, 852 w, 814 w, 784 w, 756 w, 668 w, 644 w .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(500 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): 1.92(d, J=1.3, \mathrm{Me}-\mathrm{C}(5))$; $2.16\left(d, J=13.5,8.0,1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.17\left(d d, J=13.0,10.0,1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 2.34\left(d d, J=13.4,11.8,1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.42$ $\left(d d, J=13.0,5.2,1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 4.23-4.25\left(m, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right), \mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 4.55\left(d d d, J=11.4,8.4,2.6, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right) ; 6.47$ $\left(d d, J=10.1,5.2, \mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 8.19(d, J=1.3, \mathrm{H}-\mathrm{C}(6)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(125 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): 12.79(q, M e-\mathrm{C}(5))$; $42.43,47.65\left(2 t, \mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(7^{\prime}\right)\right) ; 54.65\left(d, \mathrm{C}\left(6^{\prime}\right)\right) ; 72.86\left(d, \mathrm{C}\left(4^{\prime}\right)\right.$ or $\left.\mathrm{C}\left(5^{\prime}\right)\right) ; 85.71\left(s, \mathrm{C}\left(3^{\prime}\right)\right) ; 89.37\left(d, \mathrm{C}\left(1^{\prime}\right)\right) ; 92.52$ $\left(d, \mathrm{C}\left(4^{\prime}\right)\right.$ or $\left.\mathrm{C}\left(5^{\prime}\right)\right) ; 111.83(s, \mathrm{C}(5)) ; 117.69\left(q, J(\mathrm{C}, \mathrm{F})=287, \mathrm{CF}_{3}\right) ; 138.93(d, \mathrm{C}(6)) ; 152.60(s, \mathrm{C}(2)) ; 158.77$ $\left(q, J(\mathrm{C}, \mathrm{F})=37, \mathrm{COCF}_{3}\right) ; 166.66(s, \mathrm{C}(4))$. LSI-MS: $380.0\left(18,[M+\mathrm{H}]^{+}\right), 127(100)$.
( $3^{\prime} \mathrm{S}, 5^{\prime} \mathrm{R}, 6^{\prime} \mathrm{R}$ )-1-( $6^{\prime}$-Amino-2'-deoxy-3', $5^{\prime}$-ethano- $\beta$-D-ribofuranosyl)thymine ( $\mathbf{1 3}$ ). A soln. of $\mathbf{1 2}$ (20 mg, $53 \mu \mathrm{~mol})$ in $\mathrm{MeOH}(0.1 \mathrm{ml})$ was treated with conc. $\mathrm{NH}_{3}$ soln. $(1.5 \mathrm{ml})$. After 45 min at r.t., the mixture was lyophilized and the residue adsorbed on silica gel $(\mathrm{MeOH})$ and purified by $\mathrm{FC}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 5: 1+5 \%\right.$ conc. $\mathrm{NH}_{3}$ soln. ): $\mathbf{1 3}(15 \mathrm{mg}, 98 \%)$. Colorless film. TLC $\left(\mathrm{MeCl}_{2} / \mathrm{MeOH} 4: 1+5 \%\right.$ conc. $\left.\mathrm{NH}_{3}\right): R_{\mathrm{f}} 0.25$. UV ( MeOH ): 267 (9770). IR (KBr): $3072 m$ (br.), 2963m, 1684s, 1472m, 1436w, 1292m, 1264m, 1203s, 1132s, 1046m, 934w. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CD}_{3} \mathrm{OD}\right): 1.92(d, J=1.1, \mathrm{Me}-\mathrm{C}(5)) ; 2.18-2.37\left(m, 1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right), 2 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.44$ $\left(d d, J=13.1,5.4,1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 3.86\left(d d d, J=9.8,8.7,3.7, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right) ; 4.21\left(d, J=4.2, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right) ; 4.31(t, J=4.0$, $\left.\mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 6.40\left(d d, J=9.9,5.1, \mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 7.96(d, J=1.1, \mathrm{H}-\mathrm{C}(6)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 12.8$ $(q, M e-\mathrm{C}(5)) ; 42.5,47.4,\left(2 t, \mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(7^{\prime}\right)\right) ; 54.6\left(d, \mathrm{C}\left(6^{\prime}\right)\right) ; 72.0\left(d, \mathrm{C}\left(5^{\prime}\right)\right) ; 85.9\left(s, \mathrm{C}\left(3^{\prime}\right)\right) ; 90.0,92.2\left(2 d, \mathrm{C}\left(1^{\prime}\right)\right.$, $\left.\mathrm{C}\left(4^{\prime}\right)\right) ; 112.1(s, \mathrm{C}(5)) ; 139.0(d, \mathrm{C}(6)) ; 152.7(s, \mathrm{C}(2)) ; 166.6(s, \mathrm{C}(4))$. HR-LSI-MS $\left(\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{~N}_{3} \mathrm{O}_{5}\right): 284.1245$ (calc. 284.1246).
( $\left.3^{\prime} \mathrm{S}, 5^{\prime} \mathrm{R}, 6^{\prime} \mathrm{R}\right)$-1-\{2'-Deoxy-5'-O-[(4,4'-dimethoxytriphenyl)methyl]-6'-(trifluoroacetamido)-3', $5^{\prime}$-ethano- $\beta$-D-ribofuranosylfthymine (14). A soln. of $\mathbf{1 2}(173 \mathrm{mg}, 0.46 \mathrm{mmol})$ in dry pyridine $(0.5 \mathrm{ml})$, containing activated molecular sieves ( $3 \AA$ ), was treated with $(\mathrm{MeO})_{2} \mathrm{TrOSO}_{2} \mathrm{CF}_{3}(312 \mathrm{mg}, 0.69 \mathrm{mmol})$ and heated to $50^{\circ}$. After 9 h , another portion of $(\mathrm{MeO})_{2} \mathrm{TrOSO}_{2} \mathrm{CF}_{3}(103 \mathrm{mg}, 0.23 \mathrm{mmol})$ was added. After a total of 16 h , the mixture was diluted with $\operatorname{AcOEt}(30 \mathrm{ml})$ and extracted with sat. $\mathrm{NaHCO}_{3}$ soln. $(30 \mathrm{ml})$, the org. phase dried $\left(\mathrm{MgSO}_{4}\right)$ and evaporated, and the residual oil purified by $\mathrm{FC}\left(2.5-10 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}+1 \% \mathrm{Et}_{3} \mathrm{~N}\right): \mathbf{1 4}(144 \mathrm{mg}, 46 \%)$, besides $40 \%$ of recovered starting material. TLC $\left(8 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}\right): R_{\mathrm{f}} 0.46$. IR $\left(\mathrm{CHCl}_{3}\right): 3595 w, 3393 w$, $2961 w, 2935 w, 2912 w, 2839 w, 1694 s, 1628 w, 1608 m, 1580 w, 1528 w, 1512 m, 1467 m, 1443 w, 1426 w, 1413 w, 1370 w$, $1320 w, 1303 m, 1281 m, 1257 s, 1237 s, 1216 s, 1201 \mathrm{vs}, 1180 s, 1097 m, 1057 m, 1035 m, 1014 m, 910 w, 872 w, 825 m$. ${ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 1.76\left(d d, J=15.3,6.4,1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 1.93(s, \mathrm{Me}-\mathrm{C}(5)) ; 2.04(d d, J=13.6,9.9$, $\left.1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 2.35\left(d, J=15.1,1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.49\left(d d, J=13.8,5.3,1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 3.00\left(d d, J=10.5,5.7, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right)$; 3.4 (br. $s, \mathrm{OH}) ; 3.75(s, \mathrm{MeO}) ; 3.89\left(d, J=5.5, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right) ; 4.22\left(t, J=6.1, \mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 6.19(d d, J=9.6,5.5$, $\left.\mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 6.79-6.83(m, 4$ arom. H) ; 7.16-7.50(m, 14 H$) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 12.16(q, M e-\mathrm{C}(5))$; $42.13,47.92\left(2 t, \mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(7^{\prime}\right)\right) ; 53.89\left(d, \mathrm{C}\left(6^{\prime}\right)\right) ; 55.29(q, \mathrm{MeO}) ; 72.29\left(d, \mathrm{C}\left(5^{\prime}\right)\right) ; 83.47,88.10\left(2 s, \mathrm{C}\left(3^{\prime}\right)\right.$, $\left.\mathrm{Ar}_{2} C \mathrm{Ph}\right) ; 84.85\left(d, \mathrm{C}\left(1^{\prime}\right)\right) ; 88.69\left(d, \mathrm{C}\left(4^{\prime}\right)\right) ; 115.72\left(q, J(\mathrm{C}, \mathrm{F})=289, \mathrm{CF}_{3}\right) ; 112.36(s, \mathrm{C}(5)) ; 113.57$, 113.61, $127.41,127.53,128.23,129.56,129.67$ ( $7 d$, arom. C); 134.56 ( $d, \mathrm{C}(6)$ ); 135.30, $135.59,144.47$ ( $3 s$, arom. C); $150.52(s, \mathrm{C}(2)) ; 156.76\left(q, J(\mathrm{C}, \mathrm{F})=37, \mathrm{COCF}_{3}\right) ; 159.04,159.07(2 s$, arom. C $) ; 163.39(s, \mathrm{C}(4))$. LSI-MS: 720.1 $\left(2,[M+\mathrm{K}]^{+}\right), 681.1\left(2, M^{+}\right), 303.1(100)$.
( $3^{\prime} \mathrm{S}, 5^{\prime} \mathrm{R}, 6^{\prime} \mathrm{R}$ )-1-\{ $6^{\prime}-$ Acetamido-2'-deoxy-5'-O-[(4,4'-dimethoxytriphenyl)methyl]-3',5'-ethano- $\beta$-D-ribofuranosylfthymine (15). As described for $\mathbf{1 4}$, from $11(186 \mathrm{mg}, 0.57 \mathrm{mmol})$ and $(\mathrm{MeO})_{2} \mathrm{TrOSO}_{2} \mathrm{CF}_{3}$ ( 395 mg , $0.87 \mathrm{mmol})$ in pyridine $(0.6 \mathrm{ml})$. $\mathrm{FC}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 50: 1\right)$ gave $15(277 \mathrm{mg}, 77 \%)$. Slightly brownish foam. TLC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 50: 4\right): R_{\mathrm{f}} 0.53 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 1.70\left(d d, J=14.8,6.3,1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 1.90$ $(d, J=0.9, \mathrm{Me}-\mathrm{C}(5)) ; 1.94(s, \mathrm{MeCO}) ; 2.21-2.32\left(m, 1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right), 1 \mathrm{H}-\mathrm{C}\left(7^{\prime}\right)\right) ; 2.41(d d, J=13.7,5.8$, $\left.1 \mathrm{H}-\mathrm{C}\left(2^{\prime}\right)\right) ; 3.26$ (br. $\left.s, \mathrm{OH}-\mathrm{C}\left(3^{\prime}\right)\right) ; 3.32\left(d d, J=11.0,6.0, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right) ; 3.65\left(d, J=5.5, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right) ; 3.74$ $(s, \mathrm{MeO}) ; 4.08-4.14\left(m, \mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 6.01\left(d d, J=9.4,5.9, \mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 6.55(d, J=4.6, \mathrm{CONH}) ; 6.79(d, J=8.6$, 4 arom. H$) ; 7.21(d, J=0.9, \mathrm{H}-\mathrm{C}(6)) ; 7.16-7.50(3 m, 9$ arom. H$) ; 9.10(s, \mathrm{H}-\mathrm{N}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}(75 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right): 12.61(q, M e-\mathrm{C}(5)) ; 23.40(q, M e \mathrm{CO}) ; 43.33,47.41\left(2 t, \mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(7^{\prime}\right)\right) ; 53.66\left(d, \mathrm{C}\left(6^{\prime}\right)\right) ; 55.25(q, \mathrm{MeO})$; $72.94\left(d, \mathrm{C}\left(5^{\prime}\right)\right) ; 83.35,87.94\left(2 s, \mathrm{C}\left(3^{\prime}\right), \mathrm{Ar}_{2} C \mathrm{Ph}\right) ; 85.64\left(d, \mathrm{C}\left(1^{\prime}\right)\right) ; 88.78\left(d, \mathrm{C}\left(4^{\prime}\right)\right) ; 111.40(s, \mathrm{C}(5)) ; 113.40$, $127.21,127.85,128.06,129.84$ ( $5 d$, arom. C) ; $135.82(d, \mathrm{C}(6)) ; 135.94,144.84$ ( $2 s$, arom. C); 150.41 ( $s, \mathrm{C}(2)$ ); 158.83 ( $s$, arom. C); 163.57 ( $s, \mathrm{C}(4))$; $170.41(s, C O)$. LSI-MS: $628\left(3,[M+1]^{+}\right), 303(100)$. Anal. calc. for $\mathrm{C}_{35} \mathrm{H}_{37} \mathrm{~N}_{3} \mathrm{O}_{8} \cdot 0.5 \mathrm{H}_{2} \mathrm{O}: \mathrm{C} 66.03, \mathrm{H} 6.02$, N 6.60 ; found: $\mathrm{C} 66.24, \mathrm{H} 6.25, \mathrm{~N} 6.56$.
( $\left.3^{\prime} \mathrm{S}, 5^{\prime} \mathrm{R}, 6^{\prime} \mathrm{R}\right)$-1-\{2'-Deoxy-5'-O-[(4,4'-dimethoxytriphenyl)methyl]- $6^{\prime}$-(trifluoroacetamido) $\mathbf{3}^{\prime}, 5^{\prime}$-ethano- $\beta$-Dribofuranosyllthymine 3'-(2-Cyanoethyl Diisopropylphosphoramidite) (16). A soln. of $\mathbf{1 4}$ ( $73 \mathrm{mg}, 0.11 \mathrm{mmol}$ ) in $\mathrm{MeCN}(1 \mathrm{ml})$ was treated with ${ }^{\mathrm{i}} \mathrm{Pr}_{2} \mathrm{NEt}(73.5 \mu \mathrm{l}, 0.43 \mathrm{mmol})$ and $\left[\mathrm{P}\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CN}\right)\left(\mathrm{N}^{\mathrm{i}} \mathrm{Pr}_{2}\right)\right] \mathrm{Cl}(48 \mu \mathrm{l}, 0.22 \mathrm{mmol})$ and the resulting mixture stirred for 2 h at r.t. Then additional ${ }^{i} \mathrm{Pr}_{2} \mathrm{NEt}(37 \mu \mathrm{l}, 0.22 \mathrm{mmol})$ and
$\mathrm{Cl}\left[\mathrm{P}\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CN}\right)\left(\mathrm{N}^{i} \mathrm{Pr}_{2}\right)\right](24 \mu \mathrm{l}, 0.11 \mathrm{mmol})$ were added, and the mixture was worked up after 7 h by dilution with AcOEt and extraction with sat. $\mathrm{NaHCO}_{3}$ soln. After drying and evaporation of the org. phase, the residual gum was purified by FC (hexane/AcOEt $2: 1 \rightarrow 0: 1$ ): $\mathbf{1 6}(83 \mathrm{mg}, 88 \%)$. White foam. TLC (hexane/ AcOEt 1:1): $R_{\mathrm{f}} 0.31,0.26 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 1.05-1.13\left(m, 2 \mathrm{Me}_{2} \mathrm{CH}\right) ; 1.93,1.94(2 d, J=0.7$, $\mathrm{Me}-\mathrm{C}(5)) ; 2.02-2.20\left(m, 2 \mathrm{H}, \mathrm{CH}_{2}\left(2^{\prime}\right)\right.$ or $\left.\mathrm{CH}_{2}\left(7^{\prime}\right)\right) ; 2.34-2.39\left(m, 1 \mathrm{H}, \mathrm{CH}_{2}\left(2^{\prime}\right)\right.$ or $\left.\mathrm{CH}_{2}\left(7^{\prime}\right)\right) ; 2.50-2.59$ $\left(m, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CN}\right) ; 2.84-2.92\left(m, 1 \mathrm{H}, \quad \mathrm{CH}_{2}\left(2^{\prime}\right) \quad\right.$ or $\left.\mathrm{CH}_{2}\left(7^{\prime}\right)\right) ; 3.17-3.27\left(m, \mathrm{Me}_{2} \mathrm{CH}\right) ; 3.44-3.65$ $\left(m, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CN}\right) ; 3.65-3.75\left(m, \mathrm{H}-\mathrm{C}\left(5^{\prime}\right)\right) ; 3.76(s, \mathrm{MeO}) ; 3.82,3.99\left(2 d, J=5.3, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right) ; 4.11-4.19$ $\left(m, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right) ; 5.96-6.04\left(m, \mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 6.80-6.84(m, 4$ arom. H$) ; 7.12-7.50(m, \mathrm{H}-\mathrm{C}(6), 9$ arom. H$) ; 9.21$ (br. $s, \mathrm{H}-\mathrm{N}(3)) .{ }^{13} \mathrm{C}-\mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 12.18(q, M e-\mathrm{C}(5)) ; 20.10,20.13,20.20,20.23(4 t, J(\mathrm{C}, \mathrm{P})=2.4$, $\left.\mathrm{NCCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right)$; 24.16, 24.19, 24.27, 24.29, 24.37, 24.40, 24.46, 24.48 ( $\left.8 q, M e_{2} \mathrm{CH}\right) ; 41.56,41.917(2 t, J(\mathrm{C}, \mathrm{P})=8.9$, $\left.\mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(7^{\prime}\right)\right)$; 43.27, $43.43\left(2 d, \mathrm{Me}_{2} C \mathrm{H}\right) ; 44.97,45.81\left(2 t, J(\mathrm{C}, \mathrm{P})=9.0, \mathrm{C}\left(2^{\prime}\right), \mathrm{C}\left(7^{\prime}\right)\right) ; 53.20,53.43\left(2 d, \mathrm{C}\left(6^{\prime}\right)\right)$; $55.22(q, \mathrm{MeO}) ; 57.66,57.89\left(2 t, J(\mathrm{C}, \mathrm{P})=9.5, \quad \mathrm{NCCH}_{2} \mathrm{CH}_{2} \mathrm{O}\right) ; 72.12,72.30\left(2 d, \mathrm{C}\left(5^{\prime}\right)\right) ; 85.55,86.15$ $\left(2 d, J(\mathrm{C}, \mathrm{P})=1.2, \mathrm{C}\left(1^{\prime}\right)\right) ; 86.99,87.23\left(2 s, J(\mathrm{C}, \mathrm{P})=7.9,8.5, \mathrm{C}\left(3^{\prime}\right)\right) ; 87.46,87.69\left(2 d, J(\mathrm{C}, \mathrm{P})=5.5,7.3, \mathrm{C}\left(4^{\prime}\right)\right)$; 88.06 ( $\left.s, \mathrm{C}\left(3^{\prime}\right), \mathrm{Ar}_{2} C \mathrm{Ph}\right) ; 111.90,111.91$ ( $2 s, \mathrm{C}(5)$ ); 113.47, 113.50, 113.53 ( $3 d$, arom. C ); 113.81, 117.64 $\left(2 q, J(\mathrm{C}, \mathrm{P})=288, \mathrm{CF}_{3}\right) ; 117.52,117.55(2 s, \mathrm{CN}) ; 127.26,127.29,127.54,127.57,128.14,128.17,129.59,129.63$, $129.69,129.71$ ( $10 d$, arom. C); 135.04, $135.06,135.55,135.58$ ( $4 d, \mathrm{C}(6)$, arom. C); $135.34,135.42,144.49,144.52$ ( $4 s$, arom. C); $150.29,150.38(2 s, \mathrm{C}(2)) ; 156.72\left(q, J(\mathrm{C}, \mathrm{P})=37, \mathrm{COCF}_{3}\right) ; 158.93,158.95,158.97$ ( $3 s$, arom. C ); $163.50(s, \mathrm{C}(4)) \cdot{ }^{31} \mathrm{P}-\mathrm{NMR}\left(81 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 141.6,142.6$.
(3'S, $\left.5^{\prime} \mathrm{R}, 6^{\prime} \mathrm{R}\right)$-1-\{( $6^{\prime}-$ Acetamido-2'-deoxy-5'-O-[(4,4'-dimethoxytriphenyl)methyl]-3',5'-ethano- $\beta$-D-ribofuranosyllthymine 3'-(2-Cyanoethyl Diisopropylphosphoramidite) (17). As described for 16, from 15 ( 213 mg , $0.34 \mathrm{mmol}),{ }^{i} \mathrm{Pr}_{2} \mathrm{NEt}(235 \mu \mathrm{l}, 1.37 \mathrm{mmol})$, and $\left[\mathrm{P}\left(\mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CN}\right)\left(\mathrm{N}^{\mathrm{i}} \mathrm{Pr}_{2}\right)\right] \mathrm{Cl}(155 \mu \mathrm{l}, 0.69 \mathrm{mmol})$ in MeCN (3 $\mathrm{ml})$. $\mathrm{FC}\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 50: 1\right)$ gave $17(176 \mathrm{mg}, 62 \% ; c a .1: 1$ diastereoisomer mixture $)$. White foam. TLC $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 25: 2\right): R_{\mathrm{f}} 0.48,0.65 .{ }^{1} \mathrm{H}-\mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 1.04-1.12\left(m, 2 \mathrm{Me}{ }_{2} \mathrm{CH}\right) ; 1.92$, 1.93 ( $2 s$, $\mathrm{Me}-\mathrm{C}(5)) ; 1.95,1.96(2 s, 3 \mathrm{MeCO}) ; 1.96-2.02\left(m, 1 \mathrm{H}, \mathrm{CH}_{2}\left(2^{\prime}\right)\right.$ or $\left.\mathrm{CH}_{2}\left(7^{\prime}\right)\right) ; 2.34-2.39\left(m, 2 \mathrm{H}, \mathrm{CH}_{2}\left(2^{\prime}\right)\right.$ or $\left.\mathrm{CH}_{2}\left(7^{\prime}\right)\right) ; 2.49-2.57\left(m, \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CN}\right) ; 2.77-2.83\left(m, 1 \mathrm{H}, \mathrm{CH}_{2}\left(2^{\prime}\right)\right.$ or $\left.\mathrm{CH}_{2}\left(7^{\prime}\right)\right) ; 3.41-3.75\left(m, \mathrm{H}-\mathrm{C}\left(4^{\prime}\right)\right.$, $\left.\mathrm{H}-\mathrm{C}\left(5^{\prime}\right), \mathrm{OCH}_{2} \mathrm{CH}_{2} \mathrm{CN}, 2 \mathrm{Me}_{2} \mathrm{CH}_{2}\right) ; 3.77(s, 2 \mathrm{MeO}) ; 4.01-4.11\left(m, \mathrm{H}-\mathrm{C}\left(6^{\prime}\right)\right) ; 5.83-5.93\left(m, \mathrm{H}-\mathrm{C}\left(1^{\prime}\right)\right) ; 6.57$, $6.71(2 d, J=4.8,5.3, \mathrm{CONH}) ; 6.79-6.83$ ( $m, 4 \mathrm{H}$ arom.) ; 7.18-7.50 ( $m, \mathrm{H}-\mathrm{C}(6), 9$ arom. H ); 8.41 (br. $s$, $\mathrm{H}-\mathrm{N}(3)) .{ }^{31} \mathrm{P}-\mathrm{NMR}\left(81 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 141.8,142.4$. LSI-MS: $828\left(2,[M+1]^{+}\right), 303(100)$. Anal. calc. for $\mathrm{C}_{44} \mathrm{H}_{54} \mathrm{~N}_{5} \mathrm{O}_{9} \mathrm{P}: \mathrm{C} 63.83, \mathrm{H} 6.57, \mathrm{~N} 8.46$; found: C 63.53, H 6.76, N 8.37.

Oligonucleotide Synthesis. Oligonucleotide synthesis was performed on a Pharmacia Gene Assembler Special connected to a Compaq-Pro-Linea-3/25-zs personal computer. All syntheses were performed using the $1.3-\mu \mathrm{mol}$ cycle with coupling times of $6-9 \mathrm{~min}$ and detritylation times of $60-90 \mathrm{~s}$ per unnatural building block. Solvents and solns. were prepared according to the manufacturer's protocol. Phosphoramidite ( 0.1 m in MeCN ) and $1 H$-tetrazole $(0.45 \mathrm{~m}$ in MeCN$)$ solns. were equal in concentration to those used for the synthesis of natural oligodeoxynucleotides. For the synthesis of 21 and 22, the activator $1 H$-tetrazole was replaced by 5-

Table 2. Synthesis and Analytical Data of Oligonucleotides 18-22

| Sequence ( $1.3 \mu \mathrm{~mol}$ ) | HPLC | $\begin{aligned} & \text { Isolated yield } \\ & O D(260 \mathrm{~nm}) \\ & ([\%]) \end{aligned}$ | MALDI-TOF-MS$[M-\mathrm{H}]^{-}$ |  |
| :---: | :---: | :---: | :---: | :---: |
|  |  |  | $m / z$ (calc.) | $\mathrm{m} / \mathrm{z}$ (found) |
| $\mathbf{1 8} \mathrm{d}\left(\right.$ TTTTT t ${ }^{+}$TTTT $)$ | $\begin{aligned} & \text { DEAE } \mathrm{A}^{\mathrm{a}}: 20-35 \% B \text { in } 20 \mathrm{~min} ; t_{\mathrm{R}} 16.5 \\ & \left.R P^{\mathrm{b}}\right): 15-22 \% B \text { in } 30 \mathrm{~min} ; t_{\mathrm{R}} 11 \end{aligned}$ | 37 (40) | 3020.1 | 3018.6 |
| $19 \mathrm{~d}\left(\mathrm{TTTTt}^{+} \mathrm{t}^{+}\right.$TTTT $)$ | DEAE ${ }^{\mathrm{a}}$ ): $20-35 \% B$ in 20 min ; $t_{\mathrm{R}} 13$ $\left.R P^{\mathrm{b}}\right): 15-22 \% B$ in $30 \mathrm{~min} ; t_{\mathrm{R}} 9.5$ | 15 (17) | 3061.1 | 3059.7 |
| $20 \mathrm{~d}\left(\mathrm{t}^{+} \mathrm{T} \mathrm{t}^{+} \mathrm{T} \mathrm{t}^{+} \mathrm{T} \mathrm{t}^{+} \mathrm{T} \mathrm{t}^{+} \mathrm{T}\right)$ | DEAE $^{\text {a }}$ ): $10-18 \% B$ in $13 \mathrm{~min} ; t_{\mathrm{R}} 9$ $\left.R P^{\mathrm{b}}\right): 10-12 \% B$ in $30 \mathrm{~min} ; t_{\mathrm{R}} 11.8$ | 31 (33) | 3184.3 | 3185.9 |
| $21 \mathrm{~d}\left(\right.$ TTTTTt $^{\text {Ac }}$ TTTT $)$ | $\left.\mathrm{DEAE}^{\mathrm{a}}\right): 25-45 \% B$ in $13 \mathrm{~min} ; t_{\mathrm{R}} 18.1$ $\left.R P^{\mathrm{b}}\right): 10-30 \% B$ in $30 \mathrm{~min} ; t_{\mathrm{R}} 12.5$ | 45 (48) | 3061.9 | 3061.1 |
| $22 \mathrm{~d}\left(\mathrm{t}^{\mathrm{Ac}} \mathrm{Tt}^{\mathrm{Ac}} \mathrm{Tt}^{\text {Ac }} \mathrm{Tt}^{\text {Ac }} \mathrm{Tt}^{\text {Ac }} \mathrm{T}\right)$ | DEAE $^{\mathrm{a}}$ ): $25-55 \% B$ in $30 \mathrm{~min} ; t_{\mathrm{R}} 17.8$ $R P^{\mathrm{b}}$ ): $5-20 \% B$ in $30 \mathrm{~min} ; t_{\mathrm{R}} 21.5$ | 48 (51) | 3394.2 | 3394.1 |

a) Nucleogen DEAE 60-7, $125 \times 4.0 \mathrm{~mm}$ (Macherey \& Nagel); A: $20 \mathrm{~mm} \mathrm{KH}_{2} \mathrm{PO}_{4}$ in $\mathrm{H}_{2} \mathrm{O} / \mathrm{MeCN} 4: 1, \mathrm{pH} 6.0 ; B: A+1 \mathrm{~m}$ KCl ; flow $1 \mathrm{ml} / \mathrm{min}$; detection at $260 \mathrm{~nm} .{ }^{\mathrm{b}}$ ) Aquapore Rp-300 $220 \times 4.6 \mathrm{~mm}, 7 \mu \mathrm{~m}$ (Brownlee Labs); A: 0.1m $\left(\mathrm{Et}_{3} \mathrm{NH}\right) \mathrm{OAc}$ in $\mathrm{H}_{2} \mathrm{O}, \mathrm{pH} 7.0 ; B: 0.1 \mathrm{~m}\left(\mathrm{Et}_{3} \mathrm{NH}\right) \mathrm{OAc}$ in $\mathrm{H}_{2} \mathrm{O} / \mathrm{MeCN} 1: 4$, pH 7.0; flow $1 \mathrm{ml} / \mathrm{min}$; detection at 260 nm .
(benzylthio)- $1 H$-tetrazole ( 0.25 m in MeCN ) [24]. Average coupling yields, monitored by the on-line trityl assay, were in the range of $90-98 \%$ for $\mathbf{1 8}$ and $\mathbf{2 0}-\mathbf{2 2}$. For sequence $\mathbf{1 9}$, the coupling step between the two adjacent trifluoroacetamido-bicyclo-thymidine residues proceeded with only $59 \%$ yield. All syntheses were run in the trityl-off mode.

Deprotection and Purification of Oligonucleotides. Removal of the protecting groups and detachment from the solid support was effected in conc. $\mathrm{NH}_{3}$ soln. $(1-2 \mathrm{ml})$ at r.t. for $13-18 \mathrm{~h}$. The crude oligomers were purified by DEAE ion-exchange HPLC, desalted over Sep-Pak (Waters), and their purity controlled by reversed-phase chromatography. Table 2 contains synthetic and anal. data of the oligonucleotides described here. All natural DNA sequences used in this study were prepared according to standard CED- or PAC-phosphoramidite chemistry and purified by HPLC.

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[^0]:    1) Part 6: [1].
[^1]:    ${ }^{2}$ ) The relative configurations at the centers $\mathrm{C}(7)$ and $\mathrm{C}(8)$ of all intermediates $\mathbf{2}-\mathbf{7}$ were assigned on the basis of ${ }^{1} \mathrm{H}-\mathrm{NMR}$ NOE experiments and are discussed in detail in [14].

[^2]:    ${ }^{6}$ ) Support for this interpretation comes from additional equilibration experiments. When $\beta-\mathbf{8}$ was subjected to the reaction conditions for nucleosidation at room temperature, equilibration to a mixture $\alpha / \beta-81.5: 1$ was observed after 23 h . No such equilibration was observed when pure $\alpha-\mathbf{8}$ was used.

[^3]:    ${ }^{7}$ ) All complexes formed between the oligonucleotides $\mathbf{1 8} \mathbf{- 2 3}$ and $\mathrm{d}\left(\mathrm{A}_{10}\right)$ at $1: 1$ stoichiometry of single strands were duplexes and not triple helices. This is deduced from the fact that only monophasic and not biphasic melting curves were observed at 260 nm , and that no transitions at 284 nm occurred. At the latter wavelength, the Hoogsteen-strand melting in $\mathrm{dT} \cdot \mathrm{dA} \cdot \mathrm{dT}$ triple helices is visible.

